

# 2-Piperidinone

<b>Other names:</b>	2-Piperidone «alpha»-Piperidone «delta»-Valerolactam Pentanoic acid, 5-amino-, lactam 5-Pentanolactam Piperidon Piperidone-2 Delta-valerolactam NSC 18894 NSC 2305 Valerolactim
<b>Inchi:</b>	InChI=1S/C5H9NO/c7-5-3-1-2-4-6-5/h1-4H2,(H,6,7)
<b>InchiKey:</b>	XUWHAWMETYGRKB-UHFFFAOYSA-N
<b>Formula:</b>	C5H9NO
<b>SMILES:</b>	O=C1CCCCN1
<b>Mol. weight [g/mol]:</b>	99.13
<b>CAS:</b>	675-20-7

## Physical Properties

Property code	Value	Unit	Source
chs	-2947.20 ± 0.42	kJ/mol	NIST Webbook
gf	-11.50	kJ/mol	Joback Method
hf	-171.76	kJ/mol	Joback Method
hfus	8.57	kJ/mol	Joback Method
hvap	38.47	kJ/mol	Joback Method
ie	9.30	eV	NIST Webbook
ie	9.15 ± 0.02	eV	NIST Webbook
log10ws	-0.77		Crippen Method
logp	0.287		Crippen Method
mcvol	82.000	ml/mol	McGowan Method
pc	5102.04	kPa	Joback Method
rinpol	1173.50		NIST Webbook
rinpol	1173.50		NIST Webbook
ripol	2060.00		NIST Webbook
ss	165.00	J/mol×K	NIST Webbook
ss	164.80	J/mol×K	NIST Webbook
tb	529.20	K	NIST Webbook

tc	692.75	K	Joback Method
tf	312.20 ± 0.50	K	NIST Webbook
tt	311.85 ± 0.05	K	NIST Webbook
vc	0.293	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	160.54	J/mol×K	454.39	Joback Method
cpg	173.63	J/mol×K	494.12	Joback Method
cpg	230.64	J/mol×K	692.75	Joback Method
cpg	220.43	J/mol×K	653.02	Joback Method
cpg	209.61	J/mol×K	613.29	Joback Method
cpg	198.19	J/mol×K	573.57	Joback Method
cpg	186.19	J/mol×K	533.84	Joback Method
cps	144.20	J/mol×K	300.00	NIST Webbook
cps	189.80	J/mol×K	290.00	NIST Webbook
cps	208.30	J/mol×K	295.00	NIST Webbook
hfust	10.50	kJ/mol	311.90	NIST Webbook
hfust	10.49	kJ/mol	311.85	NIST Webbook
hfust	10.50	kJ/mol	311.85	NIST Webbook
hvapt	75.50	kJ/mol	302.50	NIST Webbook
sfust	33.70	J/mol×K	311.85	NIST Webbook
sfust	33.60	J/mol×K	311.85	NIST Webbook

## Sources

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C675207&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

## Legend

**chs:** Standard solid enthalpy of combustion

<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>ss:</b>	Solid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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