

# Propane, 2-isocyano-2-methyl-

<b>Other names:</b>	(CH <sub>3</sub> ) <sub>3</sub> CNC 2-Methyl-2-isocyanopropane 2-methylpropyl isocyanide TBIN t-Butyl isocyanide tert-Butyl Isocyanide tert-Butylisonitrile
<b>Inchi:</b>	InChI=1S/C5H9N/c1-5(2,3)6-4/h1-3H3
<b>InchiKey:</b>	FAGLEPBREOXSA-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>5</sub> H <sub>9</sub> N
<b>SMILES:</b>	[C-]#[N+]C(C)(C)C
<b>Mol. weight [g/mol]:</b>	83.13
<b>CAS:</b>	7188-38-7

## Physical Properties

Property code	Value	Unit	Source
affp	870.70	kJ/mol	NIST Webbook
basg	838.30	kJ/mol	NIST Webbook
gf	127.24	kJ/mol	Joback Method
hf	9.60	kJ/mol	Joback Method
hfus	2.80	kJ/mol	Joback Method
hvap	35.91	kJ/mol	Joback Method
ie	10.50	eV	NIST Webbook
log10ws	-3.79		Crippen Method
logp	1.704		Crippen Method
mcvol	82.690	ml/mol	McGowan Method
pc	3443.98	kPa	Joback Method
tb	365.00	K	NIST Webbook
tc	616.42	K	Joback Method
tf	282.15 ± 4.00	K	NIST Webbook
vc	0.331	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	155.24	J/mol×K	412.65	Joback Method
cpg	164.35	J/mol×K	446.61	Joback Method
cpg	172.92	J/mol×K	480.57	Joback Method
cpg	180.97	J/mol×K	514.54	Joback Method
cpg	188.53	J/mol×K	548.50	Joback Method
cpg	195.62	J/mol×K	582.46	Joback Method
cpg	202.27	J/mol×K	616.42	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.52593e+01
Coeff. B	-3.43148e+03
Coeff. C	-4.25230e+01
Temperature range (K), min.	271.72
Temperature range (K), max.	387.47

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7188387&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7188387&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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