

# 4-Dichloromethyl-4-methyl-2,5-cyclohexadien-1-one

<b>Other names:</b>	4-Dichloromethyl-4-methyl-2,5-cyclohexadienone 2,5-Cyclohexadien-1-one, 4-(dichloromethyl)-4-methyl- 6-Dichloromethyl-6-methyl-1,4--cyclohexadien-3-one
<b>Inchi:</b>	InChI=1S/C8H8Cl2O/c1-8(7(9)10)4-2-6(11)3-5-8/h2-5,7H,1H3
<b>InchiKey:</b>	MJRRCEHODULPAV-UHFFFAOYSA-N
<b>Formula:</b>	C8H8Cl2O
<b>SMILES:</b>	CC1(C(Cl)Cl)C=CC(=O)C=C1
<b>Mol. weight [g/mol]:</b>	191.06
<b>CAS:</b>	6611-78-5

## Physical Properties

Property code	Value	Unit	Source
chs	-4130.20	kJ/mol	NIST Webbook
gf	-53.53	kJ/mol	Joback Method
hf	-197.79	kJ/mol	Joback Method
hfus	8.84	kJ/mol	Joback Method
hvap	45.89	kJ/mol	Joback Method
log10ws	-2.72		Crippen Method
logp	2.491		Crippen Method
mcvol	130.170	ml/mol	McGowan Method
pc	3480.65	kPa	Joback Method
tb	542.79	K	Joback Method
tc	794.55	K	Joback Method
tf	325.78	K	Joback Method
vc	0.485	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	266.99	J/molxK	542.79	Joback Method
cpg	279.83	J/molxK	584.75	Joback Method
cpg	291.73	J/molxK	626.71	Joback Method
cpg	302.83	J/molxK	668.67	Joback Method
cpg	313.27	J/molxK	710.63	Joback Method

cpg	323.18	J/mol×K	752.59	Joback Method
cpg	332.70	J/mol×K	794.55	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6611785&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6611785&amp;Units=SI</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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