

# 1,2,3,6,7,8-Hexahydro-as-indacene

<b>Other names:</b>	As-Indacene, 1,2,3,6,7,8-hexahydro- 1,2,3,6,7,8-Hexahydro-cis-indacene
<b>Inchi:</b>	InChI=1S/C12H14/c1-3-9-7-8-10-4-2-6-12(10)11(9)5-1/h7-8H,1-6H2
<b>InchiKey:</b>	CGIPSWGAIQVCNY-UHFFFAOYSA-N
<b>Formula:</b>	C12H14
<b>SMILES:</b>	<chem>c1cc2c(c3c1CCC3)CCC2</chem>
<b>Mol. weight [g/mol]:</b>	158.24
<b>CAS:</b>	1076-17-1

## Physical Properties

Property code	Value	Unit	Source
gf	270.60	kJ/mol	Joback Method
hf	97.39	kJ/mol	Joback Method
hfus	13.84	kJ/mol	Joback Method
hvap	47.01	kJ/mol	Joback Method
ie	8.09	eV	NIST Webbook
ie	8.66 ± 0.02	eV	NIST Webbook
log10ws	-3.63		Crippen Method
logp	2.664		Crippen Method
mvol	134.460	ml/mol	McGowan Method
pc	3235.66	kPa	Joback Method
tb	538.40	K	Joback Method
tc	774.59	K	Joback Method
tf	333.34	K	Joback Method
vc	0.515	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	316.68	J/mol×K	538.40	Joback Method
cpg	389.15	J/mol×K	735.23	Joback Method
cpg	376.80	J/mol×K	695.86	Joback Method
cpg	363.50	J/mol×K	656.50	Joback Method
cpg	349.13	J/mol×K	617.13	Joback Method

cpg	333.56	J/mol×K	577.77	Joback Method
cpg	400.68	J/mol×K	774.59	Joback Method
dvisc	0.0007272	Paxs	538.40	Joback Method
dvisc	0.0008027	Paxs	504.22	Joback Method
dvisc	0.0008987	Paxs	470.05	Joback Method
dvisc	0.0010242	Paxs	435.87	Joback Method
dvisc	0.0011935	Paxs	401.69	Joback Method
dvisc	0.0014309	Paxs	367.52	Joback Method
dvisc	0.0017806	Paxs	333.34	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1076171&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1076171&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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