

# cis-Bicyclo[8.1.0]undecane

<b>Inchi:</b>	InChI=1S/C11H20/c1-2-4-6-8-11-9-10(11)7-5-3-1/h10-11H,1-9H2/t10-,11+
<b>InchiKey:</b>	ZGGJKSINQDYFJN-PHIMTYICSA-N
<b>Formula:</b>	C11H20
<b>SMILES:</b>	C1CCCCC2CC2CCC1
<b>Mol. weight [g/mol]:</b>	152.28

## Physical Properties

Property code	Value	Unit	Source
gf	102.74	kJ/mol	Joback Method
hf	-155.57	kJ/mol	Joback Method
hfus	10.02	kJ/mol	Joback Method
hvap	40.77	kJ/mol	Joback Method
log10ws	-3.73		Crippen Method
logp	3.757		Crippen Method
mvol	144.130	ml/mol	McGowan Method
pc	2802.44	kPa	Joback Method
rinpol	1216.00		NIST Webbook
rinpol	1221.00		NIST Webbook
tb	485.91	K	Joback Method
tc	713.20	K	Joback Method
tf	232.01	K	Joback Method
vc	0.525	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	334.24	J/mol×K	485.91	Joback Method
cpg	358.53	J/mol×K	523.79	Joback Method
cpg	381.33	J/mol×K	561.67	Joback Method
cpg	402.70	J/mol×K	599.56	Joback Method
cpg	422.69	J/mol×K	637.44	Joback Method
cpg	441.37	J/mol×K	675.32	Joback Method
cpg	458.77	J/mol×K	713.20	Joback Method
dvisc	0.0074291	Paxs	232.01	Joback Method

dvisc	0.0029653	Paxs	274.33	Joback Method
dvisc	0.0015129	Paxs	316.64	Joback Method
dvisc	0.0009046	Paxs	358.96	Joback Method
dvisc	0.0006029	Paxs	401.28	Joback Method
dvisc	0.0004341	Paxs	443.59	Joback Method
dvisc	0.0003310	Paxs	485.91	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R293426&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R293426&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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