

# 1-Phenylheneicosane

|                             |  |
|-----------------------------|--|
| <b>Other names:</b>         | 11-Phenylheneicosane<br>Benzene, heneicosyl-                                       |
| <b>Inchi:</b>               | InChI=1S/C27H48/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-15-16-17-18-19-21-24-27-25-22-20 |
| <b>InchiKey:</b>            | OWXCBZFAHWMZCQ-UHFFFAOYSA-N  |
| <b>Formula:</b>             | C27H48   |
| <b>SMILES:</b>              | CCCCCCCCCCCCCCCCCCCCc1ccccc1   |
| <b>Mol. weight [g/mol]:</b> | 372.67   |
| <b>CAS:</b>                 | 40775-09-5   |

## Physical Properties

| Property code | Value            | Unit                 | Source         |
|---------------|------------------|----------------------|----------------|
| chl           | -16977.80 ± 6.70 | kJ/mol               | NIST Webbook   |
| chl           | -16980.80 ± 6.70 | kJ/mol               | NIST Webbook   |
| gf            | 288.87           | kJ/mol               | Joback Method  |
| hf            | -364.08          | kJ/mol               | Joback Method  |
| hfl           | -506.80 ± 6.70   | kJ/mol               | NIST Webbook   |
| hfl           | -508.10 ± 6.70   | kJ/mol               | NIST Webbook   |
| hfus          | 59.73            | kJ/mol               | Joback Method  |
| hvap          | 77.97            | kJ/mol               | Joback Method  |
| log10ws       | -10.23           |                      | Crippen Method |
| logp          | 9.661            |                      | Crippen Method |
| mcvol         | 367.530          | ml/mol               | McGowan Method |
| pc            | 826.21           | kPa                  | Joback Method  |
| tb            | 843.84           | K                    | Joback Method  |
| tc            | 1034.66          | K                    | Joback Method  |
| tf            | 420.47           | K                    | Joback Method  |
| vc            | 1.440            | m <sup>3</sup> /kmol | Joback Method  |

## Temperature Dependent Properties

| Property code | Value   | Unit    | Temperature [K] | Source        |
|---------------|---------|---------|-----------------|---------------|
| cpg           | 1196.03 | J/mol×K | 843.84          | Joback Method |
| cpg           | 1218.17 | J/mol×K | 875.64          | Joback Method |
| cpg           | 1239.09 | J/mol×K | 907.45          | Joback Method |

|       |           |         |         |               |
|-------|-----------|---------|---------|---------------|
| cpg   | 1258.88   | J/mol×K | 939.25  | Joback Method |
| cpg   | 1277.60   | J/mol×K | 971.05  | Joback Method |
| cpg   | 1295.31   | J/mol×K | 1002.85 | Joback Method |
| cpg   | 1312.07   | J/mol×K | 1034.66 | Joback Method |
| dvisc | 0.0012137 | Paxs    | 420.47  | Joback Method |
| dvisc | 0.0004526 | Paxs    | 491.03  | Joback Method |
| dvisc | 0.0002163 | Paxs    | 561.59  | Joback Method |
| dvisc | 0.0001219 | Paxs    | 632.15  | Joback Method |
| dvisc | 0.0000770 | Paxs    | 702.72  | Joback Method |
| dvisc | 0.0000530 | Paxs    | 773.28  | Joback Method |
| dvisc | 0.0000388 | Paxs    | 843.84  | Joback Method |
| hvapt | 108.40    | kJ/mol  | 575.50  | NIST Webbook  |

## Correlations

| Information                 | Value                         |
|-----------------------------|-------------------------------|
| Property code               | pvap                          |
| Equation                    | $\ln(P_{vp}) = A + B/(T + C)$ |
| Coeff. A                    | 1.40815e+01                   |
| Coeff. B                    | -5.44421e+03                  |
| Coeff. C                    | -1.34916e+02                  |
| Temperature range (K), min. | 529.60                        |
| Temperature range (K), max. | 755.69                        |

## Sources

The Yaws Handbook of Vapor

Pressure:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

Joback Method:

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C40775095&Units=SI>

## Legend

chl: Standard liquid enthalpy of combustion

|                 |   |
|-----------------|---|
| <b>cpg:</b>     | Ideal gas heat capacity                                   |
| <b>dvisc:</b>   | Dynamic viscosity   |
| <b>gf:</b>      | Standard Gibbs free energy of formation                   |
| <b>hf:</b>      | Enthalpy of formation at standard conditions              |
| <b>hfl:</b>     | Liquid phase enthalpy of formation at standard conditions |
| <b>hfus:</b>    | Enthalpy of fusion at standard conditions                 |
| <b>hvap:</b>    | Enthalpy of vaporization at standard conditions           |
| <b>hvapt:</b>   | Enthalpy of vaporization at a given temperature           |
| <b>log10ws:</b> | Log10 of Water solubility in mol/l                        |
| <b>logp:</b>    | Octanol/Water partition coefficient                       |
| <b>mcvol:</b>   | McGowan's characteristic volume                           |
| <b>pc:</b>      | Critical Pressure   |
| <b>pvap:</b>    | Vapor pressure  |
| <b>tb:</b>      | Normal Boiling Point Temperature                          |
| <b>tc:</b>      | Critical Temperature                                      |
| <b>tf:</b>      | Normal melting (fusion) point                             |
| <b>vc:</b>      | Critical Volume   |

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