

# 2-Butenoic acid, ethyl ester, (Z)-

<b>Other names:</b>	Ethyl (Z)-2-butenate cis-Ethyl crotonate
<b>Inchi:</b>	InChI=1S/C6H10O2/c1-3-5-6(7)8-4-2/h3,5H,4H2,1-2H3/b5-3-
<b>InchiKey:</b>	ZFDIRQKJPRINOQ-HYXAFXHYSA-N
<b>Formula:</b>	C6H10O2
<b>SMILES:</b>	CC=CC(=O)OCC
<b>Mol. weight [g/mol]:</b>	114.14
<b>CAS:</b>	6776-19-8

## Physical Properties

Property code	Value	Unit	Source
gf	-154.06	kJ/mol	Joback Method
hf	-294.75	kJ/mol	Joback Method
hfus	14.28	kJ/mol	Joback Method
hvap	38.06	kJ/mol	Joback Method
log10ws	-1.05		Crippen Method
logp	1.126		Crippen Method
mvol	98.540	ml/mol	McGowan Method
pc	3476.55	kPa	Joback Method
rinpol	830.00		NIST Webbook
rinpol	830.00		NIST Webbook
tb	417.13	K	Joback Method
tc	603.73	K	Joback Method
tf	224.46	K	Joback Method
vc	0.376	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	184.29	J/mol×K	417.13	Joback Method
cpg	227.98	J/mol×K	572.63	Joback Method
cpg	219.98	J/mol×K	541.53	Joback Method
cpg	211.61	J/mol×K	510.43	Joback Method
cpg	202.88	J/mol×K	479.33	Joback Method

cpg	193.78	J/mol×K	448.23	Joback Method
cpg	235.63	J/mol×K	603.73	Joback Method
dvisc	0.0002255	Paxs	417.13	Joback Method
dvisc	0.0002879	Paxs	385.02	Joback Method
dvisc	0.0003840	Paxs	352.91	Joback Method
dvisc	0.0005428	Paxs	320.80	Joback Method
dvisc	0.0008286	Paxs	288.68	Joback Method
dvisc	0.0014062	Paxs	256.57	Joback Method
dvisc	0.0027763	Paxs	224.46	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C6776198&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C6776198&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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