

# Trichlormethiazide

## Other names:

2H-1,2,4-Benzothiadiazine-7-sulfonamide,  
6-chloro-3-(dichloromethyl)-3,4-dihydro-1,1-dioxide  
3-(Dichloromethyl)-6-chloro-7-sulfamoyl-3,4-dihydro-1,2,4-benzothiadiazine  
1,1-dioxide  
3-Dichloromethyl-6-chloro-7-sulfamoyl-3,4-dihydro-1,2,4-benzothiadiazine  
1,1-dioxide  
3-Dichloromethylhydrochlorothiazide  
6-Chloro-3-(dichloromethyl)-3,4-dihydro-2H-1,2,4-benzothiadiazine-7-sulfonamide  
1,1-dioxide  
6-Chloro-3-(dichloromethyl)-3,4-dihydro-2H-1,2,4-thiadiazine-7-sulfonamide  
1,1-dioxide  
6-Chloro-3-(dichloromethyl)3,4-dihydro-7-sulfamoyl-1,2,4-benzothiadiazine-1,1-dioxide  
6-Chloro-3-(dichloromethyl)3,4-dihydro-7-sulfamoyl-1,2,4-benzothiadiazine-1,1-dioxide  
Achletin  
Anatran  
Anistadin  
Aponorin  
Carvacron  
Chlopolidine  
Ciba 7057-Su  
Cretonin  
Diu-Hydrin  
Diurazida  
Diurese  
Diuroral  
Esmarin  
Eurinol  
Fluitran  
Flurese  
Flutra  
Gangesol  
Hydrotrichlorothiazide  
Intromene  
Isestran  
Kubacron  
Metahydrin  
Metatensin  
NSC 61560  
Nakva  
Naqua  
Naquasone  
Naquival  
Salurin  
Schebitran  
Tachionin

Tolcasone  
Trichlormetazid  
Trichlormethiazid  
Trichloromethiadiazide  
Trichloromethiazide  
Triclordiuride  
Triclormetiazide  
Triflumen

**Inchi:** InChI=1S/C8H8Cl3N3O4S2/c9-3-1-4-6(2-5(3)19(12,15)16)20(17,18)14-8(13-4)7(10)11/h  
**InchiKey:** LMJSLTNSBFUCMU-UHFFFAOYSA-N  
**Formula:** C8H8Cl3N3O4S2  
**SMILES:** NS(=O)(=O)c1cc2c(cc1Cl)NC(C(Cl)Cl)NS2(=O)=O  
**Mol. weight [g/mol]:** 380.66  
**CAS:** 133-67-5

## Physical Properties

Property code	Value	Unit	Source
gf	-578.05	kJ/mol	Joback Method
hf	-786.09	kJ/mol	Joback Method
hfus	61.11	kJ/mol	Joback Method
hvap	110.94	kJ/mol	Joback Method
log10ws	-2.68		Estimated Solubility Method
logp	0.821		Crippen Method
mcvol	211.800	ml/mol	McGowan Method
pc	6359.24	kPa	Joback Method
tb	791.16	K	Joback Method
tc	1034.13	K	Joback Method
tf	752.57	K	Joback Method
vc	0.812	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	511.99	J/molxK	791.16	Joback Method
cpg	521.45	J/molxK	831.65	Joback Method
cpg	529.72	J/molxK	872.15	Joback Method

cpg	536.78	J/mol×K	912.64	Joback Method
cpg	542.63	J/mol×K	953.14	Joback Method
cpg	547.27	J/mol×K	993.63	Joback Method
cpg	550.69	J/mol×K	1034.13	Joback Method

## Sources

<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C133675&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C133675&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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