

# 2H-Tetrazole, 2-methyl-

<b>Other names:</b>	2-Methyltetrazole
<b>Inchi:</b>	InChI=1S/C2H4N4/c1-6-4-2-3-5-6/h2H,1H3
<b>InchiKey:</b>	VRESBNUEIKZECD-UHFFFAOYSA-N
<b>Formula:</b>	C2H4N4
<b>SMILES:</b>	Cn1ncnn1
<b>Mol. weight [g/mol]:</b>	84.08
<b>CAS:</b>	16681-78-0

## Physical Properties

Property code	Value	Unit	Source
chl	-1640.20 ± 0.50	kJ/mol	NIST Webbook
hf	328.40 ± 0.70	kJ/mol	NIST Webbook
hfl	281.60 ± 0.60	kJ/mol	NIST Webbook
hvap	46.80	kJ/mol	NIST Webbook
hvap	46.80 ± 0.40	kJ/mol	NIST Webbook
ie	10.00	eV	NIST Webbook
log10ws	-2.06		Crippen Method
logp	-0.790		Crippen Method
mcvol	59.500	ml/mol	McGowan Method
tt	286.00 ± 1.00	K	NIST Webbook

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hfust	12.37	kJ/mol	286.00	NIST Webbook
hfust	12.37	kJ/mol	286.00	NIST Webbook

## Sources

**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C16681780&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci990307l>

**Crippen Method:**

[https://www.cheméo.com/doc/models/crippen\\_log10ws](https://www.cheméo.com/doc/models/crippen_log10ws)

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>tt:</b>	Triple Point Temperature

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