

# Fenipentol

<b>Other names:</b>	Benzenemethanol, «alpha»-butyl- Benzyl alcohol, «alpha»-butyl- «alpha»-Butylbenzenemethanol «alpha»-Butylbenzyl alcohol Pancoral Phenylbutylcarbinol 1-Pentanol, 1-phenyl- 1-Phenyl-1-pentanol 1-Phenylpentanol Suiclisin 1-Phenyl-1-hydroxypentane 1-Hydroxy-1-phenylpentane PC 1 PH BC 1-Phenyl-1-hydroxy-n-pentane NSC 8478
<b>Inchi:</b>	InChI=1S/C11H16O/c1-2-3-9-11(12)10-7-5-4-6-8-10/h4-8,11-12H,2-3,9H2,1H3
<b>InchiKey:</b>	OVGORFFCBUIFIA-UHFFFAOYSA-N
<b>Formula:</b>	C11H16O
<b>SMILES:</b>	CCCCC(O)c1ccccc1
<b>Mol. weight [g/mol]:</b>	164.24
<b>CAS:</b>	583-03-9

## Physical Properties

Property code	Value	Unit	Source
gf	14.89	kJ/mol	Joback Method
hf	-191.35	kJ/mol	Joback Method
hfus	18.85	kJ/mol	Joback Method
hvap	58.65	kJ/mol	Joback Method
log10ws	-3.25		Crippen Method
logp	2.910		Crippen Method
mcvol	147.960	ml/mol	McGowan Method
pc	2944.08	kPa	Joback Method
rinpol	1321.00		NIST Webbook
tb	569.50	K	Joback Method
tc	763.30	K	Joback Method
tf	285.97	K	Joback Method

vc

0.556

m<sup>3</sup>/kmol

Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	358.85	J/mol×K	569.50	Joback Method
cpg	372.41	J/mol×K	601.80	Joback Method
cpg	385.22	J/mol×K	634.10	Joback Method
cpg	397.31	J/mol×K	666.40	Joback Method
cpg	408.71	J/mol×K	698.70	Joback Method
cpg	419.45	J/mol×K	731.00	Joback Method
cpg	429.57	J/mol×K	763.30	Joback Method
dvisc	0.0199438	Paxs	285.97	Joback Method
dvisc	0.0042092	Paxs	333.23	Joback Method
dvisc	0.0013074	Paxs	380.48	Joback Method
dvisc	0.0005258	Paxs	427.74	Joback Method
dvisc	0.0002535	Paxs	474.99	Joback Method
dvisc	0.0001394	Paxs	522.25	Joback Method
dvisc	0.0000847	Paxs	569.50	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C583039&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C583039&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log10 of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>p<sub>c</sub>:</b>	Critical Pressure
<b>r<sub>inpol</sub>:</b>	Non-polar retention indices
<b>t<sub>b</sub>:</b>	Normal Boiling Point Temperature
<b>t<sub>c</sub>:</b>	Critical Temperature
<b>t<sub>f</sub>:</b>	Normal melting (fusion) point
<b>v<sub>c</sub>:</b>	Critical Volume

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