

# Carbonic acid, allyl 4-isopropylphenyl ester

<b>Inchi:</b>	InChI=1S/C13H16O3/c1-4-9-15-13(14)16-12-7-5-11(6-8-12)10(2)3/h4-8,10H,1,9H2,2-3H
<b>InchiKey:</b>	HGLOCQRSJASENM-UHFFFAOYSA-N
<b>Formula:</b>	C13H16O3
<b>SMILES:</b>	C=CCOC(=O)Oc1ccc(C(C)C)cc1
<b>Mol. weight [g/mol]:</b>	220.26

## Physical Properties

Property code	Value	Unit	Source
gf	-92.16	kJ/mol	Joback Method
hf	-343.46	kJ/mol	Joback Method
hfus	22.25	kJ/mol	Joback Method
hvap	57.98	kJ/mol	Joback Method
log10ws	-3.73		Crippen Method
logp	3.511		Crippen Method
mcvol	179.280	ml/mol	McGowan Method
pc	2331.52	kPa	Joback Method
rinqol	1591.00		NIST Webbook
tb	623.45	K	Joback Method
tc	833.19	K	Joback Method
tf	352.84	K	Joback Method
vc	0.672	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	448.15	J/molxK	623.45	Joback Method
cpg	463.15	J/molxK	658.41	Joback Method
cpg	477.30	J/molxK	693.36	Joback Method
cpg	490.60	J/molxK	728.32	Joback Method
cpg	503.08	J/molxK	763.28	Joback Method
cpg	514.73	J/molxK	798.23	Joback Method
cpg	525.57	J/molxK	833.19	Joback Method
dvisc	0.0014541	Paxs	352.84	Joback Method
dvisc	0.0007699	Paxs	397.94	Joback Method

dvisc	0.0004640	Paxs	443.04	Joback Method
dvisc	0.0003070	Paxs	488.15	Joback Method
dvisc	0.0002179	Paxs	533.25	Joback Method
dvisc	0.0001631	Paxs	578.35	Joback Method
dvisc	0.0001273	Paxs	623.45	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U357380&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U357380&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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