

# Diethylmalonic acid, 8-chlorooctyl heptyl ester

<b>Inchi:</b>	InChI=1S/C22H41ClO4/c1-4-7-8-12-15-18-26-20(24)22(5-2,6-3)21(25)27-19-16-13-10-9
<b>InchiKey:</b>	VUJOTVLZAGTTDX-UHFFFAOYSA-N
<b>Formula:</b>	C22H41ClO4
<b>SMILES:</b>	CCCCCCCOC(=O)C(CC)(CC)C(=O)OCCCCCCCCCI
<b>Mol. weight [g/mol]:</b>	405.01

## Physical Properties

Property code	Value	Unit	Source
gf	-342.57	kJ/mol	Joback Method
hf	-1011.50	kJ/mol	Joback Method
hfus	55.09	kJ/mol	Joback Method
hvap	85.97	kJ/mol	Joback Method
log10ws	-6.67		Crippen Method
logp	6.429		Crippen Method
mcvol	347.960	ml/mol	McGowan Method
pc	943.26	kPa	Joback Method
rinpol	2570.00		NIST Webbook
rinpol	2570.00		NIST Webbook
tb	889.54	K	Joback Method
tc	1089.21	K	Joback Method
tf	514.36	K	Joback Method
vc	1.353	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	1123.76	J/molxK	889.54	Joback Method
cpg	1202.42	J/molxK	1055.93	Joback Method
cpg	1188.91	J/molxK	1022.65	Joback Method
cpg	1174.33	J/molxK	989.37	Joback Method
cpg	1158.65	J/molxK	956.10	Joback Method
cpg	1141.80	J/molxK	922.82	Joback Method
cpg	1214.92	J/molxK	1089.21	Joback Method
dvisc	0.0000262	Paxs	889.54	Joback Method

dvisc	0.0000354	Paxs	827.01	Joback Method
dvisc	0.0000502	Paxs	764.48	Joback Method
dvisc	0.0000757	Paxs	701.95	Joback Method
dvisc	0.0001238	Paxs	639.42	Joback Method
dvisc	0.0002252	Paxs	576.89	Joback Method
dvisc	0.0004739	Paxs	514.36	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U370753&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U370753&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/22-384-1/Diethylmalonic-acid-8-chlorooctyl-heptyl-ester.pdf>

Generated by Cheméo on 2024-04-18 05:23:15.244218313 +0000 UTC m=+15707044.164795634.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.