

# Glutaric acid, hexyl pentachlorophenyl ester

<b>Inchi:</b>	InChI=1S/C17H19Cl5O4/c1-2-3-4-5-9-25-10(23)7-6-8-11(24)26-17-15(21)13(19)12(18)14
<b>InchiKey:</b>	LJWJFVLDUHIUJV-UHFFFAOYSA-N
<b>Formula:</b>	C17H19Cl5O4
<b>SMILES:</b>	CCCCCOC(=O)CCCC(=O)Oc1c(Cl)c(Cl)c(Cl)c(Cl)c1Cl
<b>Mol. weight [g/mol]:</b>	464.60

## Physical Properties

Property code	Value	Unit	Source
gf	-370.97	kJ/mol	Joback Method
hf	-783.33	kJ/mol	Joback Method
hfus	58.44	kJ/mol	Joback Method
hvap	99.26	kJ/mol	Joback Method
log10ws	-7.84		Crippen Method
logp	7.153		Crippen Method
mcvol	302.710	ml/mol	McGowan Method
pc	1385.05	kPa	Joback Method
rinqol	3018.00		NIST Webbook
tb	979.67	K	Joback Method
tc	1206.48	K	Joback Method
tf	664.29	K	Joback Method
vc	1.173	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	813.93	J/molxK	979.67	Joback Method
cpg	823.16	J/molxK	1017.47	Joback Method
cpg	831.22	J/molxK	1055.27	Joback Method
cpg	838.11	J/molxK	1093.08	Joback Method
cpg	843.83	J/molxK	1130.88	Joback Method
cpg	848.38	J/molxK	1168.68	Joback Method
cpg	851.78	J/molxK	1206.48	Joback Method
dvisc	0.0002031	Paxs	664.29	Joback Method
dvisc	0.0001414	Paxs	716.85	Joback Method

dvisc	0.0001034	Paxs	769.42	Joback Method
dvisc	0.0000787	Paxs	821.98	Joback Method
dvisc	0.0000619	Paxs	874.54	Joback Method
dvisc	0.0000501	Paxs	927.11	Joback Method
dvisc	0.0000414	Paxs	979.67	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U360256&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U360256&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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