

# Benzene, azido-

<b>Other names:</b>	Azidobenzene Benzene, triazo- Phenyl azide
<b>Inchi:</b>	InChI=1S/C6H5N3/c7-9-8-6-4-2-1-3-5-6/h1-5H
<b>InchiKey:</b>	CTRLRINCMYICJO-UHFFFAOYSA-N
<b>Formula:</b>	C6H5N3
<b>SMILES:</b>	[N-]=[N+]=Nc1ccccc1
<b>Mol. weight [g/mol]:</b>	119.12
<b>CAS:</b>	622-37-7

## Physical Properties

Property code	Value	Unit	Source
affp	820.00	kJ/mol	NIST Webbook
basg	787.50	kJ/mol	NIST Webbook
chl	-3420.00	kJ/mol	NIST Webbook
hvap	44.80 ± 0.80	kJ/mol	NIST Webbook
ie	8.40	eV	NIST Webbook
ie	8.72 ± 0.02	eV	NIST Webbook
log10ws	-7.13		Crippen Method
logp	2.628		Crippen Method
mcpol	92.980	ml/mol	McGowan Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hvapt	45.20	kJ/mol	358.00	NIST Webbook

## Sources

**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C622377&Units=SI>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>chl:</b>	Standard liquid enthalpy of combustion
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume

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