

# 1,3,7-Octatriene

<b>Other names:</b>	«alpha»-Ocimene octa-1,3,7-triene
<b>Inchi:</b>	InChI=1S/C8H12/c1-3-5-7-8-6-4-2/h3-5,7H,1-2,6,8H2
<b>InchiKey:</b>	ZTJHDEXGCKAXRZ-UHFFFAOYSA-N
<b>Formula:</b>	C8H12
<b>SMILES:</b>	C=CC=CCCC=C
<b>Mol. weight [g/mol]:</b>	108.18
<b>CAS:</b>	1002-35-3

## Physical Properties

Property code	Value	Unit	Source
gf	272.38	kJ/mol	Joback Method
hf	159.63	kJ/mol	Joback Method
hfus	14.12	kJ/mol	Joback Method
hvap	32.02	kJ/mol	Joback Method
log10ws	-2.73		Crippen Method
logp	2.695		Crippen Method
mcvol	110.680	ml/mol	McGowan Method
pc	2921.84	kPa	Joback Method
rinpol	808.00		NIST Webbook
rinpol	805.00		NIST Webbook
rinpol	805.00		NIST Webbook
ripol	1249.00		NIST Webbook
tb	393.00 ± 4.00	K	NIST Webbook
tc	559.15	K	Joback Method
tf	171.32	K	Joback Method
vc	0.425	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	189.08	J/mol×K	379.96	Joback Method
cpg	242.22	J/mol×K	529.28	Joback Method
cpg	232.69	J/mol×K	499.42	Joback Method

cpg	222.63	J/molxK	469.55	Joback Method
cpg	212.03	J/molxK	439.69	Joback Method
cpg	200.86	J/molxK	409.82	Joback Method
cpg	251.26	J/molxK	559.15	Joback Method
dvisc	0.0001901	Paxs	379.96	Joback Method
dvisc	0.0002419	Paxs	345.19	Joback Method
dvisc	0.0003250	Paxs	310.41	Joback Method
dvisc	0.0004703	Paxs	275.64	Joback Method
dvisc	0.0007574	Paxs	240.87	Joback Method
dvisc	0.0014324	Paxs	206.09	Joback Method
dvisc	0.0035085	Paxs	171.32	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C1002353&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C1002353&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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