

# Benzenemethanol, 4-chloro-«alpha»-phenyl-

<b>Other names:</b>	Benzhydrol, 4-chloro- (4-Chlorophenyl)phenylmethanol p-Chlorobenzhydrol Benzhydrol, p-chloro- Chlorobenzhydrol 4-Chlorobenzhydrol 4-Chlorodiphenyl carbinol 4-Chlorobenzohydrol NSC 59990 4-chlorobenzhydryl alcohol
<b>Inchi:</b>	InChI=1S/C13H11ClO/c14-12-8-6-11(7-9-12)13(15)10-4-2-1-3-5-10/h1-9,13,15H
<b>InchiKey:</b>	AJYOOHCNOXWTKJ-UHFFFAOYSA-N
<b>Formula:</b>	C13H11ClO
<b>SMILES:</b>	OC(c1ccccc1)c1ccc(Cl)cc1
<b>Mol. weight [g/mol]:</b>	218.68
<b>CAS:</b>	119-56-2

## Physical Properties

Property code	Value	Unit	Source
gf	122.58	kJ/mol	Joback Method
hf	-23.31	kJ/mol	Joback Method
hfus	21.88	kJ/mol	Joback Method
hvap	70.42	kJ/mol	Joback Method
log10ws	-3.98		Crippen Method
logp	3.422		Crippen Method
mcvol	164.620	ml/mol	McGowan Method
pc	3243.03	kPa	Joback Method
rinpol	2335.00		NIST Webbook
rinpol	2335.00		NIST Webbook
ripol	3383.00		NIST Webbook
ripol	3383.00		NIST Webbook
tb	684.35	K	Joback Method
tc	916.51	K	Joback Method
tf	377.37	K	Joback Method
vc	0.610	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	402.13	J/molxK	684.35	Joback Method
cpg	454.21	J/molxK	877.82	Joback Method
cpg	445.48	J/molxK	839.13	Joback Method
cpg	435.97	J/molxK	800.43	Joback Method
cpg	425.62	J/molxK	761.74	Joback Method
cpg	414.36	J/molxK	723.04	Joback Method
cpg	462.22	J/molxK	916.51	Joback Method
dvisc	0.0000455	Paxs	684.35	Joback Method
dvisc	0.0000687	Paxs	633.19	Joback Method
dvisc	0.0001113	Paxs	582.02	Joback Method
dvisc	0.0001980	Paxs	530.86	Joback Method
dvisc	0.0003983	Paxs	479.70	Joback Method
dvisc	0.0009468	Paxs	428.53	Joback Method
dvisc	0.0028462	Paxs	377.37	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C119562&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C119562&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient

<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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