

# Pentasulfide, dimethyl

<b>Other names:</b>	Dimethyl pentasulfide Dimethyl pentasulphide
<b>Inchi:</b>	InChI=1S/C2H6S5/c1-3-5-7-6-4-2/h1-2H3
<b>InchiKey:</b>	HFHOSCCMQJJVKR-UHFFFAOYSA-N
<b>Formula:</b>	C2H6S5
<b>SMILES:</b>	CSSSSSC
<b>Mol. weight [g/mol]:</b>	190.39
<b>CAS:</b>	7330-31-6

## Physical Properties

Property code	Value	Unit	Source
gf	131.56	kJ/mol	Joback Method
hf	124.74	kJ/mol	Joback Method
hfus	21.59	kJ/mol	Joback Method
hvap	54.13	kJ/mol	Joback Method
log10ws	-4.04		Crippen Method
logp	3.572		Crippen Method
mcvol	120.790	ml/mol	McGowan Method
pc	5695.98	kPa	Joback Method
rinpol	1426.00		NIST Webbook
rinpol	1421.00		NIST Webbook
rinpol	1428.00		NIST Webbook
rinpol	1415.00		NIST Webbook
rinpol	1396.00		NIST Webbook
rinpol	1415.00		NIST Webbook
rinpol	1426.00		NIST Webbook
rinpol	1421.00		NIST Webbook
rinpol	1428.00		NIST Webbook
rinpol	1428.00		NIST Webbook
rinpol	1396.00		NIST Webbook
ripol	2185.00		NIST Webbook
ripol	2185.00		NIST Webbook
tb	589.06	K	Joback Method
tc	891.25	K	Joback Method
tf	284.30	K	Joback Method
vc	0.417	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	211.35	J/mol×K	589.06	Joback Method
cpg	219.72	J/mol×K	639.42	Joback Method
cpg	227.47	J/mol×K	689.79	Joback Method
cpg	234.48	J/mol×K	740.15	Joback Method
cpg	240.67	J/mol×K	790.52	Joback Method
cpg	245.94	J/mol×K	840.88	Joback Method
cpg	250.20	J/mol×K	891.25	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C7330316&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C7330316&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307I">http://pubs.acs.org/doi/abs/10.1021/ci990307I</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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