

# Cyclopropane, propyl-

<b>Other names:</b>	Propane, 1-cyclopropyl- Propylcyclopropane
<b>Inchi:</b>	InChI=1S/C6H12/c1-2-3-6-4-5-6/h6H,2-5H2,1H3
<b>InchiKey:</b>	MWVPQZRIWVPJCA-UHFFFAOYSA-N
<b>Formula:</b>	C6H12
<b>SMILES:</b>	CCCC1CC1
<b>Mol. weight [g/mol]:</b>	84.16
<b>CAS:</b>	2415-72-7

## Physical Properties

Property code	Value	Unit	Source
chl	-3770.00	kJ/mol	NIST Webbook
gf	60.39	kJ/mol	Joback Method
hf	-94.37	kJ/mol	Joback Method
hfus	9.43	kJ/mol	Joback Method
hvap	28.86	kJ/mol	Joback Method
log10ws	-1.99		Crippen Method
logp	2.196		Crippen Method
mcvol	84.540	ml/mol	McGowan Method
pc	3611.55	kPa	Joback Method
rinpol	614.80		NIST Webbook
rinpol	609.00		NIST Webbook
rinpol	608.90		NIST Webbook
rinpol	615.00		NIST Webbook
rinpol	609.00		NIST Webbook
rinpol	609.00		NIST Webbook
tb	342.30 ± 0.50	K	NIST Webbook
tb	342.65 ± 1.50	K	NIST Webbook
tb	341.65 ± 1.00	K	NIST Webbook
tc	521.85	K	Joback Method
tf	175.32	K	Joback Method
vc	0.329	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	140.88	J/molxK	343.42	Joback Method
cpg	196.23	J/molxK	492.11	Joback Method
cpg	186.28	J/molxK	462.37	Joback Method
cpg	175.79	J/molxK	432.63	Joback Method
cpg	164.75	J/molxK	402.90	Joback Method
cpg	153.12	J/molxK	373.16	Joback Method
cpg	205.68	J/molxK	521.85	Joback Method
dvisc	0.0002790	Paxs	343.42	Joback Method
dvisc	0.0003062	Paxs	315.40	Joback Method
dvisc	0.0003422	Paxs	287.39	Joback Method
dvisc	0.0003917	Paxs	259.37	Joback Method
dvisc	0.0004633	Paxs	231.35	Joback Method
dvisc	0.0005740	Paxs	203.34	Joback Method
dvisc	0.0007614	Paxs	175.32	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.52680e+01
Coeff. B	-3.25869e+03
Coeff. C	-3.63090e+01
Temperature range (K), min.	253.84
Temperature range (K), max.	363.60

## Sources

McGowan Method:

<http://link.springer.com/article/10.1007/BF02311772>

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C2415727&Units=SI>

The Yaws Handbook of Vapor Pressure:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**KDB:**

<https://www.cheric.org/files/research/kdb/mol/mol455.mol>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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