

2-Dodecene, 2,6-dimethyl

Inchi:	InChI=1S/C14H28/c1-5-6-7-8-11-14(4)12-9-10-13(2)3/h10,14H,5-9,11-12H2,1-4H3
InchiKey:	RQILLHDCUHGIST-UHFFFAOYSA-N
Formula:	C14H28
SMILES:	CCCCCCC(C)CCC=C(C)C
Mol. weight [g/mol]:	196.37

Physical Properties

Property code	Value	Unit	Source
gf	136.23	kJ/mol	Joback Method
hf	-230.14	kJ/mol	Joback Method
hfus	27.39	kJ/mol	Joback Method
hvap	46.41	kJ/mol	Joback Method
log10ws	-5.29		Crippen Method
logp	5.339		Crippen Method
mvol	203.820	ml/mol	McGowan Method
pc	1603.85	kPa	Joback Method
rinpol	1325.00		NIST Webbook
rinpol	1325.00		NIST Webbook
tb	523.32	K	Joback Method
tc	695.07	K	Joback Method
tf	213.50	K	Joback Method
vc	0.794	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	488.89	J/mol×K	523.32	Joback Method
cpg	507.42	J/mol×K	551.95	Joback Method
cpg	525.15	J/mol×K	580.57	Joback Method
cpg	542.11	J/mol×K	609.20	Joback Method
cpg	558.32	J/mol×K	637.82	Joback Method
cpg	573.82	J/mol×K	666.45	Joback Method
cpg	588.63	J/mol×K	695.07	Joback Method

Sources

Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=R47166&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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