Sucrose

Other names: (+)-Sucrose

(2R,3R,4S,5S,6R)-2-(((2S,3S,4S,5R)-3,4-dihydroxy-2,5-bis(hydroxymethyl)tetrahydrofura

(«alpha»-D-Glucosido)-«beta»-D-fructofuranoside

(«alpha»-D-Glucosido)-«beta»-D-fructofuranoside

.alpha.-trehalose

.beta.-D-fructofuranosyl .alpha.-D-glucopyranoside

4-O-.beta.-D-galactopyranosyl-D-glucose

Amerfond

Beet sugar

Cane sugar

Confectioner's sugar

D-(+)-Saccharose

D-(+)-Sucrose

D-(+)-lactose

D-Sucrose

D-trehalose

Fructofuranoside, «alpha»-D-glucopyranosyl, «beta»-D

Fructofuranoside, «alpha»-D-glucopyranosyl, «beta»-D

Glucopyranoside, «beta»-D-fructofuranosyl, «alpha»-D

Glucopyranoside, «beta»-D-fructofuranosyl, «alpha»-D

Granulated sugar

Microse

NCI-C56597

NSC 406942

Rock candy

Saccharose

Saccharum

Sugar

Table sugar

White sugar

alpha, alpha-trehalose

lactose

«alpha»-D-Glucopyranoside, «beta»-D-fructofuranosyl

«alpha»-D-Glucopyranosyl «beta»-D-fructofuranoside

«beta»-D-Fructofuranoside, «alpha»-D-glucopyranosyl

«beta»-D-Fructofuranosyl «alpha»-D-glucopyranoside

«alpha»-D-Glucopyranoside, «beta»-D-fructofuranosyl

«alpha»-D-Glucopyranosyl «beta»-D-fructofuranoside

«beta»-D-Fructofuranoside, «alpha»-D-glucopyranosyl

«beta»-D-Fructofuranosyl «alpha»-D-glucopyranoside

InChi=1S/C12H22O11/c13-1-4-6(16)8(18)9(19)11(21-4)23-12(3-15)10(20)7(17)5(2-14)23

InchiKey: CZMRCDWAGMRECN-SFOFJGFUSA-N

Formula: C12H22O11

 $\textbf{SMILES:} \qquad \qquad \mathsf{OCC1OC}(\mathsf{OC2}(\mathsf{CO})\mathsf{OC}(\mathsf{CO})\mathsf{C}(\mathsf{O})\mathsf{C2O})\mathsf{C}(\mathsf{O})\mathsf{C1O}$

Mol. weight [g/mol]: 342.30 CAS: 57-50-1

Physical Properties

Property code	Value	Unit	Source
chs	-5643.40 ± 1.80	kJ/mol	NIST Webbook
chs	-5644.17	kJ/mol	NIST Webbook
chs	-5637.40 ± 1.70	kJ/mol	NIST Webbook
chs	-5664.38 ± 0.69	kJ/mol	NIST Webbook
gf	-1320.10	kJ/mol	Joback Method
hf	-1917.41	kJ/mol	Joback Method
hfs	-2221.20	kJ/mol	NIST Webbook
hfus	63.65	kJ/mol	Joback Method
hvap	184.54	kJ/mol	Joback Method
log10ws	0.79		Aqueous Solubility Prediction Method
log10ws	0.79		Estimated Solubility Method
logp	-5.396		Crippen Method
mcvol	222.790	ml/mol	McGowan Method
рс	4627.70	kPa	Joback Method
SS	392.40	J/mol×K	NIST Webbook
SS	360.20	J/mol×K	NIST Webbook
tb	1290.10	K	Joback Method
tc	1782.75	K	Joback Method
tf	462.00 ± 3.00	K	NIST Webbook
tf	424.40	К	Heat capacity and transition behavior of sucrose by standard, fast scanning and temperature-modulated calorimetry
tf	458.65	K	Aqueous Solubility Prediction Method
tf	464.05	К	Artificial neural networks as a supporting tool for compatibility study based on thermogravimetric data
tf	461.00 ± 6.00	K	NIST Webbook
VC	0.784	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source	
cpg	970.71	J/mol×K	1290.10	Joback Method	
cpg	998.85	J/mol×K	1372.21	Joback Method	
срд	1028.30	J/mol×K	1454.32	Joback Method	
cpg	1060.01	J/mol×K	1536.42	Joback Method	
cpg	1094.94	J/mol×K	1618.53	Joback Method	
cpg	1134.04	J/mol×K	1700.64	Joback Method	
cpg	1178.26	J/mol×K	1782.75	Joback Method	
cps	451.00	J/mol×K	318.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides	
cps	416.60	J/mol×K	293.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides	
cps	424.30	J/mol×K	298.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides	
cps	429.40	J/mol×K	303.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides	
cps	437.50	J/mol×K	308.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides	
cps	445.50	J/mol×K	313.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides	

_	cps	408.50	J/mol×K	288.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides	
	cps	466.20	J/mol×K	323.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides	
	cps	472.60	J/mol×K	328.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides	
	cps	482.50	J/mol×K	333.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides	
_	cps	490.30	J/mol×K	338.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides	
	cps	498.80	J/mol×K	343.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides	
	cps	506.70	J/mol×K	348.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides	
	cps	513.90	J/mol×K	353.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides	
	cps	522.00	J/mol×K	358.15	Temperature dependence of the heat capacities in the solid state of 18 mono-, di-, and poly-saccharides	

cps	424.30	J/mol×K	298.15	NIST Webbook	
cps	430.00	J/mol×K	300.00	NIST Webbook	
cps	425.50	J/mol×K	298.15	NIST Webbook	
cps	422.50	J/mol×K	297.00	NIST Webbook	
hfust	46.20	kJ/mol	459.00	NIST Webbook	

Sources

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Legend

chs: Standard solid enthalpy of combustion

cpg: Ideal gas heat capacitycps: Solid phase heat capacity

gf: Standard Gibbs free energy of formationhf: Enthalpy of formation at standard conditions

hfs: Solid phase enthalpy of formation at standard conditions

hfus: Enthalpy of fusion at standard conditions hfust: Enthalpy of fusion at a given temperature

hvap: Enthalpy of vaporization at standard conditions

log10ws: Log10 of Water solubility in mol/llogp: Octanol/Water partition coefficientmcvol: McGowan's characteristic volume

pc: Critical Pressure

ss: Solid phase molar entropy at standard conditions

tb: Normal Boiling Point Temperature

tc: Critical Temperature

tf: Normal melting (fusion) point

vc: Critical Volume

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