

# Trichloroacetic acid isopropyl ester

<b>Other names:</b>	Acetic acid, trichloro-, 1-methylethyl ester Acetic acid, trichloro-, isopropyl ester Isopropyl trichloroacetate 1-Methylethyl trichloroacetate Trichloroacetic acid, isopropyl ester
<b>Inchi:</b>	InChI=1S/C5H7Cl3O2/c1-3(2)10-4(9)5(6,7)8/h3H,1-2H3
<b>InchiKey:</b>	JYXIYFNAIFVCAN-UHFFFAOYSA-N
<b>Formula:</b>	C5H7Cl3O2
<b>SMILES:</b>	CC(C)OC(=O)C(Cl)(Cl)Cl
<b>Mol. weight [g/mol]:</b>	205.47
<b>CAS:</b>	3974-99-0

## Physical Properties

Property code	Value	Unit	Source
chl	-2492.00	kJ/mol	NIST Webbook
chl	-2500.90 ± 4.20	kJ/mol	NIST Webbook
gf	-278.09	kJ/mol	Joback Method
hf	-484.50 ± 9.60	kJ/mol	NIST Webbook
hfl	-536.40 ± 8.40	kJ/mol	NIST Webbook
hfus	13.15	kJ/mol	Joback Method
hvap	51.90 ± 4.20	kJ/mol	NIST Webbook
log10ws	-2.45		Crippen Method
logp	2.308		Crippen Method
mcvol	125.470	ml/mol	McGowan Method
pc	3318.18	kPa	Joback Method
rinpol	995.90		NIST Webbook
rinpol	995.00		NIST Webbook
rinpol	995.00		NIST Webbook
rinpol	995.00		NIST Webbook
ripol	1312.00		NIST Webbook
ripol	1312.00		NIST Webbook
ripol	1312.00		NIST Webbook
tb	498.71	K	Joback Method
tc	714.64	K	Joback Method
tf	295.45	K	Joback Method
vc	0.469	m <sup>3</sup> /kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	234.95	J/molxK	498.71	Joback Method
cpg	243.58	J/molxK	534.70	Joback Method
cpg	251.62	J/molxK	570.69	Joback Method
cpg	259.12	J/molxK	606.67	Joback Method
cpg	266.09	J/molxK	642.66	Joback Method
cpg	272.55	J/molxK	678.65	Joback Method
cpg	278.53	J/molxK	714.64	Joback Method
dvisc	0.0043997	Paxs	295.45	Joback Method
dvisc	0.0022377	Paxs	329.33	Joback Method
dvisc	0.0012911	Paxs	363.20	Joback Method
dvisc	0.0008182	Paxs	397.08	Joback Method
dvisc	0.0005571	Paxs	430.96	Joback Method
dvisc	0.0004011	Paxs	464.83	Joback Method
dvisc	0.0003020	Paxs	498.71	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3974990&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3974990&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfl:</b>	Liquid phase enthalpy of formation at standard conditions

<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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