

# Valine

<b>Other names:</b>	L-Valine Valine, L- Butanoic acid, 2-amino-3-methyl- Butanoic acid, 2-amino-3-methyl-, (S)- L-(+)-«alpha»-Aminoisovaleric acid 2-Amino-3-methylbutanoic acid 2-Amino-3-methylbutyric acid L-iso-C <sub>3</sub> H <sub>7</sub> CH(NH <sub>2</sub> )COOH (S)-2-Amino-3-methylbutanoic acid (S)-2-Amino-3-methylbutyric acid (S)-Valine NSC 76038
<b>Inchi:</b>	InChI=1S/C <sub>5</sub> H <sub>11</sub> NO <sub>2</sub> /c1-3(2)4(6)5(7)8/h3-4H,6H <sub>2</sub> ,1-2H <sub>3</sub> ,(H,7,8)/t4-m/s1
<b>InchiKey:</b>	KZSNJWFQEVHDMF-SCSAIBSYSA-N
<b>Formula:</b>	C <sub>5</sub> H <sub>11</sub> NO <sub>2</sub>
<b>SMILES:</b>	CC(C)C(N)C(=O)O
<b>Mol. weight [g/mol]:</b>	117.15
<b>CAS:</b>	72-18-4

## Physical Properties

Property code	Value	Unit	Source
affp	910.60	kJ/mol	NIST Webbook
basg	873.00 ± 3.00	kJ/mol	NIST Webbook
basg	876.70	kJ/mol	NIST Webbook
chs	-2910.70 ± 1.90	kJ/mol	NIST Webbook
chs	-2920.10 ± 0.54	kJ/mol	NIST Webbook
gf	-212.95	kJ/mol	Joback Method
hf	-388.11	kJ/mol	Joback Method
hfs	-618.00	kJ/mol	NIST Webbook
hfs	-628.90 ± 1.90	kJ/mol	NIST Webbook
hfus	12.54	kJ/mol	Joback Method
hvap	60.01	kJ/mol	Joback Method
ie	8.71	eV	NIST Webbook
log10ws	-0.32		Crippen Method
logp	0.054		Crippen Method
mcvol	98.730	ml/mol	McGowan Method
pc	4627.70	kPa	Joback Method

ss	178.87	J/mol×K	NIST Webbook
tb	531.50	K	Joback Method
tc	722.61	K	Joback Method
tf	310.12	K	Joback Method
vc	0.357	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	264.58	J/mol×K	658.91	Joback Method
cpg	271.63	J/mol×K	690.76	Joback Method
cpg	232.36	J/mol×K	531.50	Joback Method
cpg	241.05	J/mol×K	563.35	Joback Method
cpg	249.30	J/mol×K	595.20	Joback Method
cpg	257.14	J/mol×K	627.06	Joback Method
cpg	278.30	J/mol×K	722.61	Joback Method
cps	168.50	J/mol×K	298.15	NIST Webbook
cps	168.82	J/mol×K	298.15	NIST Webbook
hsubt	163.00 ± 0.80	kJ/mol	455.00	NIST Webbook
hsubt	163.00 ± 8.00	kJ/mol	455.00	NIST Webbook

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C72184&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C72184&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity

<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>ss:</b>	Solid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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