

# 4-n-Pentanoyl-4-n'-butanoyloxyazobenzene

Inchi:	InChI=1S/C21H24N2O3/c1-3-5-7-20(24)16-8-10-17(11-9-16)22-23-18-12-14-19(15-13-18)
InchiKey:	IGGPWVPSOWNDOT-GHVJWSGMSA-N
Formula:	C21H24N2O3
SMILES:	CCCCC(=O)c1ccc(N=Nc2ccc(OC(=O)CCC)cc2)cc1
Mol. weight [g/mol]:	352.43
CAS:	120122-98-7

## Physical Properties

Property code	Value	Unit	Source
hf	-336.81	kJ/mol	Joback Method
hvap	90.79	kJ/mol	Joback Method
log10ws	-6.58		Crippen Method
logp	6.180		Crippen Method
mcvol	283.900	ml/mol	McGowan Method
pc	1317.52	kPa	Joback Method
tb	1022.56	K	Joback Method
tc	1265.36	K	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
hfust	14.23	kJ/mol	373.30	NIST Webbook
sfust	38.12	J/mol×K	373.30	NIST Webbook

## Sources

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C120122987&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Crippen Method:**

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

# Legend

<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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