

# 2H-Pyran-2-one, 6-hexyltetrahydro-

<b>Other names:</b>	.delta.-hexylvalerolactone .delta.-undecalactone .delta.-undecanolactone 2H-Pyran-2-one, tetrahydro-3-hexyl- 5-Hydroxy-n-undecanoic acid «delta»-lactone 5-Hydroxyundecanoic acid lactone 5-Undecanolide 6-hexyltetrahydro-2H-pyran-2-one Undecan-5-olide Undecanoic acid, 5-hydroxy-, «delta»-lactone Undecanolactone Undecanolide-1,5 «delta»-Hexyl-«delta»-valerolactone «delta»-Hexylvalerolactone «delta»-Undecalactone «delta»-Undecanolide
<b>Inchi:</b>	InChI=1S/C11H20O2/c1-2-3-4-5-7-10-8-6-9-11(12)13-10/h10H,2-9H2,1H3
<b>InchiKey:</b>	YZRXRLLRSPQHDK-UHFFFAOYSA-N
<b>Formula:</b>	C11H20O2
<b>SMILES:</b>	CCCCCCC1CCCC(=O)O1
<b>Mol. weight [g/mol]:</b>	184.28
<b>CAS:</b>	710-04-3

## Physical Properties

Property code	Value	Unit	Source
gf	-142.52	kJ/mol	Joback Method
hf	-485.75	kJ/mol	Joback Method
hfus	23.57	kJ/mol	Joback Method
hvap	49.27	kJ/mol	Joback Method
log10ws	-3.30		Crippen Method
logp	3.053		Crippen Method
mcvol	162.430	ml/mol	McGowan Method
pc	2395.87	kPa	Joback Method
rinpol	1579.00		NIST Webbook
rinpol	1579.00		NIST Webbook
rinpol	1535.00		NIST Webbook
rinpol	1539.00		NIST Webbook

ripol	2356.00		NIST Webbook
ripol	2345.00		NIST Webbook
ripol	2345.00		NIST Webbook
ripol	2329.00		NIST Webbook
ripol	2264.00		NIST Webbook
ripol	2336.00		NIST Webbook
sl	369.50	J/mol×K	NIST Webbook
sl	369.50	J/mol×K	NIST Webbook
tb	565.40	K	Joback Method
tc	773.82	K	Joback Method
tf	315.90	K	Joback Method
vc	0.613	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	474.35	J/mol×K	669.61	Joback Method
cpg	490.47	J/mol×K	704.34	Joback Method
cpg	505.66	J/mol×K	739.08	Joback Method
cpg	420.46	J/mol×K	565.40	Joback Method
cpg	439.35	J/mol×K	600.14	Joback Method
cpg	457.31	J/mol×K	634.87	Joback Method
cpg	519.94	J/mol×K	773.82	Joback Method
cpl	342.70	J/mol×K	298.15	NIST Webbook
cpl	342.70	J/mol×K	298.15	NIST Webbook
hvapt	80.10	kJ/mol	298.15	Vapor pressures and enthalpies of vaporization of a series of .gamma. and .delta.-lactones by correlation gas chromatography

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	426.70	K	1.50	NIST Webbook

# Sources

Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
Vapor pressures and enthalpies of vaporization of a series of $\gamma$ - and $\delta$ -lactones by correlation gas chromatography:	<a href="https://www.doi.org/10.1016/j.jct.2014.01.016">https://www.doi.org/10.1016/j.jct.2014.01.016</a>
Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C710043&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C710043&amp;Units=SI</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>cpl:</b>	Liquid phase heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>sl:</b>	Liquid phase molar entropy at standard conditions
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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