

# Furan, 2,5-dihydro-2,3,5-trimethyl

<b>Inchi:</b>	InChI=1S/C7H12O/c1-5-4-6(2)8-7(5)3/h4,6-7H,1-3H3
<b>InchiKey:</b>	RTPZOHJDYXVIFB-UHFFFAOYSA-N
<b>Formula:</b>	C7H12O
<b>SMILES:</b>	CC1=CC(C)OC1C
<b>Mol. weight [g/mol]:</b>	112.17

## Physical Properties

Property code	Value	Unit	Source
gf	-28.89	kJ/mol	Joback Method
hf	-233.36	kJ/mol	Joback Method
hfus	17.70	kJ/mol	Joback Method
hvap	36.59	kJ/mol	Joback Method
log10ws	-1.81		Crippen Method
logp	1.740		Crippen Method
mcvol	100.200	ml/mol	McGowan Method
pc	3337.38	kPa	Joback Method
rinpol	806.00		NIST Webbook
rinpol	806.00		NIST Webbook
tb	401.26	K	Joback Method
tc	599.13	K	Joback Method
tf	215.16	K	Joback Method
vc	0.374	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	193.93	J/mol×K	401.26	Joback Method
cpg	207.24	J/mol×K	434.24	Joback Method
cpg	219.97	J/mol×K	467.22	Joback Method
cpg	232.15	J/mol×K	500.20	Joback Method
cpg	243.76	J/mol×K	533.18	Joback Method
cpg	254.84	J/mol×K	566.15	Joback Method
cpg	265.39	J/mol×K	599.13	Joback Method
dvisc	0.0014047	Paxs	215.16	Joback Method

dvisc	0.0009255	Paxs	246.18	Joback Method
dvisc	0.0006694	Paxs	277.19	Joback Method
dvisc	0.0005168	Paxs	308.21	Joback Method
dvisc	0.0004183	Paxs	339.23	Joback Method
dvisc	0.0003508	Paxs	370.24	Joback Method
dvisc	0.0003023	Paxs	401.26	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R128184&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R128184&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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<https://www.chemeo.com/cid/16-297-5/Furan-2-5-dihydro-2-3-5-trimethyl.pdf>

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