

# Benzonitrile, 2-nitro-

<b>Other names:</b>	Benzonitrile, o-nitro- o-Cyanonitrobenzene o-Nitrobenzonitrile 2-Nitrobenzonitrile 2-Nitrobenzoic acid nitrile
<b>Inchi:</b>	InChI=1S/C7H4N2O2/c8-5-6-3-1-2-4-7(6)9(10)11/h1-4H
<b>InchiKey:</b>	SWBDKCMOLSUXRH-UHFFFAOYSA-N
<b>Formula:</b>	C7H4N2O2
<b>SMILES:</b>	N#Cc1cccc1[N+](=O)[O-]
<b>Mol. weight [g/mol]:</b>	148.12
<b>CAS:</b>	612-24-8

## Physical Properties

Property code	Value	Unit	Source
ea	1.98 ± 0.05	eV	NIST Webbook
ea	1.61 ± 0.10	eV	NIST Webbook
gf	279.57	kJ/mol	Joback Method
hf	191.37	kJ/mol	Joback Method
hfus	20.41	kJ/mol	Joback Method
hsub	88.10 ± 1.40	kJ/mol	NIST Webbook
hvap	61.18	kJ/mol	Joback Method
ie	10.52 ± 0.05	eV	NIST Webbook
log10ws	-2.47		Crippen Method
logp	1.466		Crippen Method
mcvol	104.530	ml/mol	McGowan Method
pc	3970.51	kPa	Joback Method
tb	645.14	K	Joback Method
tc	909.48	K	Joback Method
tf	416.19	K	Joback Method
vc	0.427	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
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cpg	262.24	J/mol×K	821.36	Joback Method
cpg	267.83	J/mol×K	865.42	Joback Method
cpg	233.38	J/mol×K	645.14	Joback Method
cpg	241.66	J/mol×K	689.20	Joback Method
cpg	249.20	J/mol×K	733.25	Joback Method
cpg	256.04	J/mol×K	777.31	Joback Method
cpg	272.86	J/mol×K	909.48	Joback Method
hfust	15.72	kJ/mol	382.70	NIST Webbook
hsubt	87.90 ± 1.40	kJ/mol	304.00	NIST Webbook

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C612248&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C612248&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hsubt:</b>	Enthalpy of sublimation at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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