

# Cyclohexanecarboxaldehyde

<b>Other names:</b>	Cyclohexanal Cyclohexanaldehyde Cyclohexanealdehyde Cyclohexylcarboxaldehyde Cyclohexylformaldehyde Formylcyclohexane 1-Formylcyclohexane Cyclohexane-1-carboxaldehyde Cyclohexanecarbaldehyde Cyclohexylmethanal Hexahydrobenzaldehyde 1-Cyclohexanecarboxaldehyde NSC 68509
<b>Inchi:</b>	InChI=1S/C7H12O/c8-6-7-4-2-1-3-5-7/h6-7H,1-5H2
<b>InchiKey:</b>	KVFDZFBHBWTVID-UHFFFAOYSA-N
<b>Formula:</b>	C7H12O
<b>SMILES:</b>	O=CC1CCCCC1
<b>Mol. weight [g/mol]:</b>	112.17
<b>CAS:</b>	2043-61-0

## Physical Properties

Property code	Value	Unit	Source
gf	-67.01	kJ/mol	Joback Method
hf	-219.07	kJ/mol	Joback Method
hfus	8.01	kJ/mol	Joback Method
hvap	38.33	kJ/mol	Joback Method
ie	9.60	eV	NIST Webbook
log10ws	-1.69		Crippen Method
logp	1.766		Crippen Method
mcvol	100.200	ml/mol	McGowan Method
pc	3853.09	kPa	Joback Method
rinpol	958.00		NIST Webbook
rinpol	963.00		NIST Webbook
rinpol	965.00		NIST Webbook
ripol	1527.00		NIST Webbook
tb	435.20	K	NIST Webbook
tb	432.50	K	NIST Webbook

tc	637.91	K	Joback Method
tf	218.03	K	Joback Method
vc	0.378	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	197.10	J/mol×K	427.77	Joback Method
cpg	212.12	J/mol×K	462.79	Joback Method
cpg	226.36	J/mol×K	497.82	Joback Method
cpg	239.83	J/mol×K	532.84	Joback Method
cpg	252.56	J/mol×K	567.87	Joback Method
cpg	264.56	J/mol×K	602.89	Joback Method
cpg	275.85	J/mol×K	637.91	Joback Method
dvisc	0.0065154	Paxs	218.03	Joback Method
dvisc	0.0028831	Paxs	252.99	Joback Method
dvisc	0.0015551	Paxs	287.94	Joback Method
dvisc	0.0009587	Paxs	322.90	Joback Method
dvisc	0.0006497	Paxs	357.86	Joback Method
dvisc	0.0004718	Paxs	392.81	Joback Method
dvisc	0.0003610	Paxs	427.77	Joback Method

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	334.50 ± 0.50	K	2.70	NIST Webbook

## Sources

- Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>
- Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)
- Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)
- McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>
- NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C2043610&Units=SI>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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