

# 3,5-Bis(trifluoromethyl)nitrobenzene

<b>Other names:</b>	3,5-di(Trifluoromethyl)nitrobenzene 5-Nitro-1,3-bis(trifluoromethyl)benzene «alpha», «alpha», «alpha», «alpha»', «alpha»', «alpha»'-Hexafluoro-5-nitro-m-xylene Benzene, 1-nitro-3,5-bis(trifluoromethyl)- «alpha», «alpha», «alpha», «alpha»', «alpha»', «alpha»'-hexafluoro-5-nitroxylene
<b>Inchi:</b>	InChI=1S/C8H3F6NO2/c9-7(10,11)4-1-5(8(12,13)14)3-6(2-4)15(16)17/h1-3H
<b>InchiKey:</b>	GMUWJDVVXLBMEZ-UHFFFAOYSA-N
<b>Formula:</b>	C8H3F6NO2
<b>SMILES:</b>	O=[N+]([O-])c1cc(C(F)(F)F)cc(C(F)(F)F)c1
<b>Mol. weight [g/mol]:</b>	259.11
<b>CAS:</b>	328-75-6

## Physical Properties

Property code	Value	Unit	Source
ea	1.79 ± 0.10	eV	NIST Webbook
gf	-1018.00	kJ/mol	Joback Method
hf	-1199.78	kJ/mol	Joback Method
hfus	24.75	kJ/mol	Joback Method
hvap	46.10	kJ/mol	Joback Method
log10ws	-4.33		Crippen Method
logp	3.632		Crippen Method
mcvol	127.860	ml/mol	McGowan Method
pc	2755.56	kPa	Joback Method
tb	560.08	K	Joback Method
tc	765.19	K	Joback Method
tf	383.37	K	Joback Method
vc	0.543	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	321.68	J/molxK	560.08	Joback Method
cpg	331.51	J/molxK	594.26	Joback Method
cpg	340.48	J/molxK	628.45	Joback Method

cpg	348.64	J/mol×K	662.63	Joback Method
cpg	356.05	J/mol×K	696.82	Joback Method
cpg	362.79	J/mol×K	731.00	Joback Method
cpg	368.90	J/mol×K	765.19	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C328756&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C328756&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>ea:</b>	Electron affinity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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