

2-Pyrimidinecarbamonitrile, 4,6-dihydroxy- (keto form)

Other names:	(1,4,5,6-tetrahydro-4,6-dioxopyrimidin-2-yl)cyanamide
Inchi:	InChI=1S/C5H4N4O2/c6-2-7-5-8-3(10)1-4(11)9-5/h1H2,(H2,7,8,9,10,11)
InchiKey:	JCHNBNRIALLIDT-UHFFFAOYSA-N
Formula:	C5H4N4O2
SMILES:	N#CNC1=NC(=O)CC(=O)N1
Mol. weight [g/mol]:	152.11
CAS:	55067-10-2

Physical Properties

Property code	Value	Unit	Source
gf	225.59	kJ/mol	Joback Method
hf	26.17	kJ/mol	Joback Method
hfus	20.66	kJ/mol	Joback Method
hvap	66.79	kJ/mol	Joback Method
log10ws	-0.26		Crippen Method
logp	-1.540		Crippen Method
mcvol	100.610	ml/mol	McGowan Method
pc	5446.55	kPa	Joback Method
tb	732.30	K	Joback Method
tc	1008.23	K	Joback Method
tf	601.67	K	Joback Method
vc	0.397	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	275.84	J/mol×K	732.30	Joback Method
cpg	286.56	J/mol×K	778.29	Joback Method
cpg	296.09	J/mol×K	824.28	Joback Method
cpg	304.27	J/mol×K	870.26	Joback Method
cpg	310.96	J/mol×K	916.25	Joback Method
cpg	316.02	J/mol×K	962.24	Joback Method
cpg	319.30	J/mol×K	1008.23	Joback Method

Sources

McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=C55067102&Units=SI
Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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