

# 2,5-Dihydroperoxy-2,5-dimethylhex-3-yne

Other names:	Hydroperoxide, 1,1'-(1,1,4,4-tetramethyl-2-butyne-1,4-diyl)bis- Hydroperoxide, (1,1,4,4-tetramethyl-2-butyne-1,4-diyl)bis-
Inchi:	InChI=1S/C8H14O4/c1-7(2,11-9)5-6-8(3,4)12-10/h9-10H,1-4H3
InchiKey:	VHRZZWMONXYKJP-UHFFFAOYSA-N
Formula:	C8H14O4
SMILES:	CC(C)(C#CC(C)(C)OO)OO
Mol. weight [g/mol]:	174.19
CAS:	3491-36-9

## Physical Properties

Property code	Value	Unit	Source
chs	-4856.80	kJ/mol	NIST Webbook
gf	-258.68	kJ/mol	Joback Method
hf	-522.55	kJ/mol	Joback Method
hfs	-257.60 ± 2.80	kJ/mol	NIST Webbook
hfus	15.32	kJ/mol	Joback Method
hsub	127.40 ± 1.80	kJ/mol	NIST Webbook
hvap	71.14	kJ/mol	Joback Method
log10ws	-1.99		Crippen Method
logp	1.526		Crippen Method
mcvol	138.460	ml/mol	McGowan Method
pc	3727.11	kPa	Joback Method
tb	614.18	K	Joback Method
tc	802.51	K	Joback Method
tf	456.96	K	Joback Method
vc	0.497	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	359.45	J/mol×K	614.18	Joback Method
cpg	369.31	J/mol×K	645.57	Joback Method
cpg	378.64	J/mol×K	676.96	Joback Method
cpg	387.44	J/mol×K	708.35	Joback Method

cpg	395.74	J/mol×K	739.73	Joback Method
cpg	403.57	J/mol×K	771.12	Joback Method
cpg	410.94	J/mol×K	802.51	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C3491369&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C3491369&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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