

# Hexethal

**Other names:**

2,4,6(1H,3H,5H)-Pyrimidinetrione, 5-ethyl-5-hexyl-  
Barbituric acid, 5-ethyl-5-hexyl-  
Ortal  
Ortol  
5-Ethyl-5-hexylbarbituric acid  
Acide hexylethylbarbiturique  
Barbituric acid, 5-ethyl-5-n-hexyl-  
5-Ethyl-5-hexyl-2,4,6(1H,3H,5H)-pyrimidinetrione  
NSC 32303

**Inchi:** InChI=1S/C12H20N2O3/c1-3-5-6-7-8-12(4-2)9(15)13-11(17)14-10(12)16/h3-8H2,1-2H3,**InchiKey:** PSTVHRSUNBSVIJ-UHFFFAOYSA-N**Formula:** C12H20N2O3**SMILES:** CCCCCC1(CC)C(=O)NC(=O)NC1=O**Mol. weight [g/mol]:** 240.30**CAS:** 77-30-5

## Physical Properties

Property code	Value	Unit	Source
gf	-123.23	kJ/mol	Joback Method
hf	-558.93	kJ/mol	Joback Method
hfus	30.08	kJ/mol	Joback Method
hvap	67.84	kJ/mol	Joback Method
log10ws	-3.18		Crippen Method
logp	1.719		Crippen Method
mcvol	193.750	ml/mol	McGowan Method
pc	2589.85	kPa	Joback Method
rinpol	1858.00		NIST Webbook
rinpol	1835.00		NIST Webbook
rinpol	1835.00		NIST Webbook
tb	794.31	K	Joback Method
tc	1037.02	K	Joback Method
tf	671.00	K	Joback Method
vc	0.734	m3/kmol	Joback Method

# Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	606.15	J/mol×K	794.31	Joback Method
cpg	624.76	J/mol×K	834.76	Joback Method
cpg	642.39	J/mol×K	875.21	Joback Method
cpg	659.04	J/mol×K	915.66	Joback Method
cpg	674.72	J/mol×K	956.12	Joback Method
cpg	689.45	J/mol×K	996.57	Joback Method
cpg	703.24	J/mol×K	1037.02	Joback Method

## Sources

Joback Method:	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
McGowan Method:	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
NIST Webbook:	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C77305&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C77305&amp;Units=SI</a>
Crippen Method:	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
Crippen Method:	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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