

# 4-Penten-1-ol

<b>Other names:</b>	4-Pentene-1-ol 4-Pentenol 4-Pentenyl alcohol CH <sub>2</sub> =CHCH <sub>2</sub> CH <sub>2</sub> CH <sub>2</sub> OH Pent-4-en-1-ol
<b>Inchi:</b>	InChI=1S/C5H10O/c1-2-3-4-5-6/h2,6H,1,3-5H2
<b>InchiKey:</b>	LQAVWYMTUMSFBE-UHFFFAOYSA-N
<b>Formula:</b>	C <sub>5</sub> H <sub>10</sub> O
<b>SMILES:</b>	C=CCCCO
<b>Mol. weight [g/mol]:</b>	86.13
<b>CAS:</b>	821-09-0

## Physical Properties

Property code	Value	Unit	Source
gf	-57.76	kJ/mol	Joback Method
hf	-173.33	kJ/mol	Joback Method
hfus	11.51	kJ/mol	Joback Method
hvap	42.73	kJ/mol	Joback Method
ie	9.42 ± 0.05	eV	NIST Webbook
log10ws	-0.15		Aqueous Solubility Prediction Method
log10ws	-0.15		Estimated Solubility Method
logp	0.945		Crippen Method
mcvol	82.880	ml/mol	McGowan Method
pc	4088.15	kPa	Joback Method
rinpol	757.00		NIST Webbook
rinpol	745.00		NIST Webbook
rinpol	739.00		NIST Webbook
rinpol	730.00		NIST Webbook
rinpol	764.00		NIST Webbook
rinpol	757.00		NIST Webbook
rinpol	739.00		NIST Webbook
rinpol	730.00		NIST Webbook
rinpol	764.00		NIST Webbook
rinpol	717.00		NIST Webbook
ripol	1297.00		NIST Webbook

ripol	1289.00		NIST Webbook
ripol	1282.00		NIST Webbook
ripol	1300.00		NIST Webbook
ripol	1299.00		NIST Webbook
ripol	1276.00		NIST Webbook
ripol	1305.00		NIST Webbook
ripol	1305.00		NIST Webbook
ripol	1308.00		NIST Webbook
ripol	1305.00		NIST Webbook
ripol	1305.00		NIST Webbook
ripol	1308.00		NIST Webbook
ripol	1295.00		NIST Webbook
ripol	1294.00		NIST Webbook
tb	408.70	K	NIST Webbook
tb	414.20	K	NIST Webbook
tb	408.00 ± 3.00	K	NIST Webbook
tb	413.00 ± 5.00	K	NIST Webbook
tc	567.49	K	Joback Method
tf	205.17	K	Joback Method
vc	0.316	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	155.12	J/molxK	402.66	Joback Method
cpg	190.73	J/molxK	540.02	Joback Method
cpg	184.19	J/molxK	512.54	Joback Method
cpg	177.36	J/molxK	485.07	Joback Method
cpg	170.25	J/molxK	457.60	Joback Method
cpg	162.84	J/molxK	430.13	Joback Method
cpg	197.00	J/molxK	567.49	Joback Method
dvisc	0.0002939	Paxs	402.66	Joback Method
dvisc	0.0005036	Paxs	369.75	Joback Method
dvisc	0.0009587	Paxs	336.83	Joback Method
dvisc	0.0020980	Paxs	303.91	Joback Method
dvisc	0.0055532	Paxs	271.00	Joback Method
dvisc	0.0192385	Paxs	238.09	Joback Method
dvisc	0.0992986	Paxs	205.17	Joback Method

# Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.59002e+01
Coeff. B	-3.99316e+03
Coeff. C	-5.47550e+01
Temperature range (K), min.	287.34
Temperature range (K), max.	431.87

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>Aqueous Solubility Prediction Method:</b>	<a href="http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa">http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDa</a>
<b>Estimated Solubility Method:</b>	<a href="http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt">http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C821090&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C821090&amp;Units=SI</a>
<b>The Yaws Handbook of Vapor Pressure:</b>	<a href="https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure">https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpolar:</b>	Non-polar retention indices

<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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