

# Anthracene, 1,2,3,4,5,6,7,8-octahydro-

<b>Other names:</b>	1,2,3,4,5,6,7,8-Octahydroanthracene Anthracene, octahydro- Octahydroanthracene s-Octahydroanthracene sym-Octahydroanthracene
<b>Inchi:</b>	InChI=1S/C14H18/c1-2-6-12-10-14-8-4-3-7-13(14)9-11(12)5-1/h9-10H,1-8H2
<b>InchiKey:</b>	LFAYMJXHGYUQNV-UHFFFAOYSA-N
<b>Formula:</b>	C14H18
<b>SMILES:</b>	<chem>c1c2c(cc3c1CCCC3)CCCC2</chem>
<b>Mol. weight [g/mol]:</b>	186.29
<b>CAS:</b>	1079-71-6

## Physical Properties

Property code	Value	Unit	Source
affp	845.40	kJ/mol	NIST Webbook
basg	814.70	kJ/mol	NIST Webbook
chs	-7962.10 ± 2.90	kJ/mol	NIST Webbook
gf	263.24	kJ/mol	Joback Method
hf	-37.20 ± 3.20	kJ/mol	NIST Webbook
hfs	-119.50 ± 2.90	kJ/mol	NIST Webbook
hfus	14.82	kJ/mol	Joback Method
hsub	82.30 ± 1.20	kJ/mol	NIST Webbook
hsub	82.30	kJ/mol	NIST Webbook
hsub	82.00 ± 1.00	kJ/mol	NIST Webbook
hvap	51.81	kJ/mol	Joback Method
ie	7.86 ± 0.05	eV	NIST Webbook
log10ws	-4.47		Crippen Method
logp	3.444		Crippen Method
mcvol	162.640	ml/mol	McGowan Method
pc	2770.08	kPa	Joback Method
rinpol	286.84		NIST Webbook
rinpol	287.07		NIST Webbook
rinpol	287.69		NIST Webbook
rinpol	287.69		NIST Webbook
rinpol	287.07		NIST Webbook
rinpol	286.84		NIST Webbook
rinpol	1673.00		NIST Webbook

rinpol	1645.00		NIST Webbook
rinpol	1667.00		NIST Webbook
rinpol	1657.00		NIST Webbook
rinpol	1694.10		NIST Webbook
rinpol	1694.10		NIST Webbook
rinpol	1680.20		NIST Webbook
rinpol	1694.10		NIST Webbook
rinpol	1703.20		NIST Webbook
rinpol	1680.20		NIST Webbook
rinpol	1694.10		NIST Webbook
rinpol	1703.20		NIST Webbook
rinpol	1694.10		NIST Webbook
rinpol	1694.10		NIST Webbook
rinpol	1704.00		NIST Webbook
rinpol	1689.00		NIST Webbook
rinpol	1684.00		NIST Webbook
ripol	2153.00		NIST Webbook
tb	567.20	K	NIST Webbook
tc	836.40	K	Joback Method
tf	347.00 ± 6.00	K	NIST Webbook
tf	346.00 ± 1.00	K	NIST Webbook
tf	347.00 ± 6.00	K	NIST Webbook
tf	346.15 ± 1.50	K	NIST Webbook
tf	346.40 ± 1.00	K	NIST Webbook
tt	345.39 ± 0.01	K	NIST Webbook
vc	0.612	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	502.17	J/mol×K	795.78	Joback Method
cpg	415.67	J/mol×K	592.70	Joback Method
cpg	435.72	J/mol×K	633.32	Joback Method
cpg	454.28	J/mol×K	673.93	Joback Method
cpg	471.46	J/mol×K	714.55	Joback Method
cpg	487.38	J/mol×K	755.17	Joback Method
cpg	515.95	J/mol×K	836.40	Joback Method
cps	327.60	J/mol×K	327.00	NIST Webbook
dvisc	0.0022319	Paxs	348.84	Joback Method
dvisc	0.0010461	Paxs	430.13	Joback Method
dvisc	0.0014688	Paxs	389.48	Joback Method

dvisc	0.0007900	Paxs	470.77	Joback Method
dvisc	0.0004287	Paxs	592.70	Joback Method
dvisc	0.0005101	Paxs	552.06	Joback Method
dvisc	0.0006239	Paxs	511.41	Joback Method
hfust	18.34	kJ/mol	345.40	NIST Webbook
hfust	2.51	kJ/mol	331.40	NIST Webbook
hfust	17.91	kJ/mol	346.00	NIST Webbook
hfust	18.34	kJ/mol	345.40	NIST Webbook
hvapt	45.60	kJ/mol	467.50	NIST Webbook
sfust	53.10	J/molxK	345.40	NIST Webbook
sfust	7.59	J/molxK	331.40	NIST Webbook
sfust	51.80	J/molxK	346.00	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	440.20	K	1.60	NIST Webbook

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	2.18648e+01
Coeff. B	-8.51461e+03
Coeff. C	-3.52500e+01
Temperature range (K), min.	429.86
Temperature range (K), max.	549.63

## Sources

NIST Webbook:

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C1079716&Units=SI>

The Yaws Handbook of Vapor Pressure:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

Crippen Method:

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

Crippen Method:

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:**

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hsub:</b>	Enthalpy of sublimation at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>hvapt:</b>	Enthalpy of vaporization at a given temperature
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>pvap:</b>	Vapor pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>sfust:</b>	Entropy of fusion at a given temperature
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>tt:</b>	Triple Point Temperature
<b>vc:</b>	Critical Volume

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