

# 4-tert-butylbenzaldehyde

<b>Inchi:</b>	InChI=1S/C11H14O/c1-11(2,3)10-6-4-9(8-12)5-7-10/h4-8H,1-3H3
<b>InchiKey:</b>	OTXINXDGSUFPNU-UHFFFAOYSA-N
<b>Formula:</b>	C11H14O
<b>SMILES:</b>	CC(C)(C)c1ccc(C=O)cc1
<b>Mol. weight [g/mol]:</b>	162.23

## Physical Properties

Property code	Value	Unit	Source
gf	47.84	kJ/mol	Joback Method
hf	-139.64	kJ/mol	Joback Method
hfus	12.77	kJ/mol	Joback Method
hvap	48.44	kJ/mol	Joback Method
log10ws	-3.03		Crippen Method
logp	2.797		Crippen Method
mcvol	143.660	ml/mol	McGowan Method
pc	2875.03	kPa	Joback Method
tb	528.17	K	Joback Method
tc	749.05	K	Joback Method
tf	297.09	K	Joback Method
vc	0.549	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	325.80	J/molxK	528.17	Joback Method
cpg	391.89	J/molxK	712.24	Joback Method
cpg	380.51	J/molxK	675.43	Joback Method
cpg	368.27	J/molxK	638.61	Joback Method
cpg	355.11	J/molxK	601.80	Joback Method
cpg	340.98	J/molxK	564.98	Joback Method
cpg	402.46	J/molxK	749.05	Joback Method
dvisc	0.0002424	Paxs	528.17	Joback Method
dvisc	0.0003151	Paxs	489.66	Joback Method
dvisc	0.0004284	Paxs	451.14	Joback Method

dvisc	0.0006168	Paxs	412.63	Joback Method
dvisc	0.0009572	Paxs	374.12	Joback Method
dvisc	0.0016432	Paxs	335.60	Joback Method
dvisc	0.0032448	Paxs	297.09	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=T999909250&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=T999909250&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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