

Ethyl furfuryl sulfide

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| Inchi: | InChI=1S/C8H12S/c1-2-9-7-8-5-3-4-6-8/h3-5H,2,6-7H2,1H3 |
| InchiKey: | PZGZDXOKXUUMRO-UHFFFAOYSA-N |
| Formula: | C8H12S |
| SMILES: | CCSCC1=CC=CC1 |
| Mol. weight [g/mol]: | 140.25 |

Physical Properties

| Property code | Value | Unit | Source |
|---------------|---------|----------------------|----------------|
| gf | 144.15 | kJ/mol | Joback Method |
| hf | 18.33 | kJ/mol | Joback Method |
| hfus | 15.53 | kJ/mol | Joback Method |
| hvap | 42.03 | kJ/mol | Joback Method |
| log10ws | -2.66 | | Crippen Method |
| logp | 2.626 | | Crippen Method |
| mcvol | 120.470 | ml/mol | McGowan Method |
| pc | 3464.28 | kPa | Joback Method |
| rinpol | 1071.00 | | NIST Webbook |
| rinpol | 1069.00 | | NIST Webbook |
| rinpol | 1067.00 | | NIST Webbook |
| rinpol | 1058.00 | | NIST Webbook |
| rinpol | 1059.00 | | NIST Webbook |
| rinpol | 1066.00 | | NIST Webbook |
| rinpol | 1057.00 | | NIST Webbook |
| rinpol | 1067.00 | | NIST Webbook |
| rinpol | 1059.00 | | NIST Webbook |
| tb | 474.47 | K | Joback Method |
| tc | 697.69 | K | Joback Method |
| tf | 243.50 | K | Joback Method |
| vc | 0.452 | m ³ /kmol | Joback Method |

Temperature Dependent Properties

| Property code | Value | Unit | Temperature [K] | Source |
|---------------|--------|---------|-----------------|---------------|
| cpg | 239.68 | J/mol×K | 474.47 | Joback Method |

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|-----|--------|---------|--------|---------------|
| cpg | 253.39 | J/mol×K | 511.67 | Joback Method |
| cpg | 266.27 | J/mol×K | 548.88 | Joback Method |
| cpg | 278.36 | J/mol×K | 586.08 | Joback Method |
| cpg | 289.70 | J/mol×K | 623.29 | Joback Method |
| cpg | 300.32 | J/mol×K | 660.49 | Joback Method |
| cpg | 310.26 | J/mol×K | 697.69 | Joback Method |

Sources

| | |
|------------------------|---|
| McGowan Method: | http://link.springer.com/article/10.1007/BF02311772 |
| NIST Webbook: | http://webbook.nist.gov/cgi/cbook.cgi?ID=R43869&Units=SI |
| Crippen Method: | http://pubs.acs.org/doi/abs/10.1021/ci9903071 |
| Crippen Method: | https://www.cheméo.com/doc/models/crippen_log10ws |
| Joback Method: | https://en.wikipedia.org/wiki/Joback_method |

Legend

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|-----------------|---|
| cpg: | Ideal gas heat capacity |
| gf: | Standard Gibbs free energy of formation |
| hf: | Enthalpy of formation at standard conditions |
| hfus: | Enthalpy of fusion at standard conditions |
| h vap: | Enthalpy of vaporization at standard conditions |
| log10ws: | Log10 of Water solubility in mol/l |
| logp: | Octanol/Water partition coefficient |
| m cvol: | McGowan's characteristic volume |
| pc: | Critical Pressure |
| rinpol: | Non-polar retention indices |
| tb: | Normal Boiling Point Temperature |
| tc: | Critical Temperature |
| tf: | Normal melting (fusion) point |
| vc: | Critical Volume |

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