Benzamide, pentafluoro-N-(pentafluorobenzoyl)-N-isobutyl-

Inchi: InChI=1S/C18H9F10NO2/c1-4(2)3-29(17(30)5-7(19)11(23)15(27)12(24)8(5)20)18(31)6-9

InchiKey: PYTMXTOCGIQRRC-UHFFFAOYSA-N

Formula: C18H9F10NO2

SMILES: CC(C)CN(C(=O)c1c(F)c(F)c(F)c(F)c1F)C(=O)c1c(F)c(F)c(F)c(F)c1F

Mol. weight [g/mol]: 461.25

Physical Properties

Property code	Value	Unit	Source
gf	-1868.40	kJ/mol	Joback Method
hf	-2180.50	kJ/mol	Joback Method
hfus	60.06	kJ/mol	Joback Method
hvap	73.81	kJ/mol	Joback Method
log10ws	-7.91		Crippen Method
logp	5.016		Crippen Method
mcvol	247.780	ml/mol	McGowan Method
pc	1320.39	kPa	Joback Method
rinpol	1658.00		NIST Webbook
rinpol	1658.00		NIST Webbook
tb	826.84	K	Joback Method
tc	1014.63	K	Joback Method
tf	593.89	K	Joback Method
vc	1.032	m3/kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	720.07	J/mol×K	826.84	Joback Method
cpg	730.37	J/mol×K	858.14	Joback Method
cpg	739.91	J/mol×K	889.44	Joback Method
cpg	748.72	J/mol×K	920.74	Joback Method
cpg	756.80	J/mol×K	952.04	Joback Method
cpg	764.18	J/mol×K	983.34	Joback Method
cpg	770.86	J/mol×K	1014.63	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci990307l
Crippen Method:	https://www.chemeo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U407954&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
hvap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
rinpol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume