

# Thiophanate

**Other names:**

Carbamic acid, [1,2-phenylenebis(iminocarbonothioyl)]bis-, diethyl ester  
Allophanic acid, 4,4'-o-phenylenebis[3-thio-, diethyl ester  
BAS 3220  
Cercobin  
Enovit  
Ethyl thiophanate  
NF 35  
NF 35, Fungicide  
PELT  
Thiofanate  
Thiophanat (German)  
Thiophanate ethyl  
Topsin  
Topsin NF 35  
1,2-Bis(3-ethoxycarbonyl-2-thioureido)benzene  
1,2-Bis[3-(ethoxycarbonyl)thioureido]benzene  
Cleary's 3336  
Diethyl 4,4'-o-phenylene-bis(3-thioallophanate)  
Thiophanat  
4,4'-o-Phenylenebis(ethyl 3-thioallophanate)  
Nemafax  
NSC 170810  
Pelt (pesticide)  
Topsin E  
Cleary 3336

**Inchi:** InChI=1S/C14H18N4O4S2/c1-3-21-13(19)17-11(23)15-9-7-5-6-8-10(9)16-12(24)18-14(20)

**InchiKey:** YFNCATAIYKQPOO-UHFFFAOYSA-N

**Formula:** C14H18N4O4S2

**SMILES:** CCOC(=O)NC(=S)Nc1ccccc1NC(=S)NC(=O)OCC

**Mol. weight [g/mol]:** 370.45

**CAS:** 23564-06-9

## Physical Properties

Property code	Value	Unit	Source
gf	293.62	kJ/mol	Joback Method
hf	-89.95	kJ/mol	Joback Method

hfus	60.84		kJ/mol	Joback Method
hvap	107.21		kJ/mol	Joback Method
log10ws	-4.75			Crippen Method
logp	2.572			Crippen Method
mcvol	263.260		ml/mol	McGowan Method
pc	2718.33		kPa	Joback Method
tb	1044.72		K	Joback Method
tc	1289.21		K	Joback Method
tf	709.98		K	Joback Method
vc	0.972		m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	783.80	J/mol×K	1044.72	Joback Method
cpg	793.56	J/mol×K	1085.47	Joback Method
cpg	802.86	J/mol×K	1126.22	Joback Method
cpg	811.82	J/mol×K	1166.96	Joback Method
cpg	820.56	J/mol×K	1207.71	Joback Method
cpg	829.19	J/mol×K	1248.46	Joback Method
cpg	837.82	J/mol×K	1289.21	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C23564069&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C23564069&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions

<b>h<sub>vap</sub>:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>w<sub>s</sub>:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>log<sub>p</sub>:</b>	Octanol/Water partition coefficient
<b>mc<sub>vol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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