

# Glutaric acid, 3-methylbut-2-yl 4-bromo-2-methoxyphenyl ester

<b>Inchi:</b>	InChI=1S/C17H23BrO5/c1-11(2)12(3)22-16(19)6-5-7-17(20)23-14-9-8-13(18)10-15(14)2
<b>InchiKey:</b>	HGUCGTFGSMQFSA-UHFFFAOYSA-N
<b>Formula:</b>	C17H23BrO5
<b>SMILES:</b>	COc1cc(Br)ccc1OC(=O)CCCC(=O)OC(C)C(C)C
<b>Mol. weight [g/mol]:</b>	387.27

## Physical Properties

Property code	Value	Unit	Source
gf	-377.99	kJ/mol	Joback Method
hf	-786.67	kJ/mol	Joback Method
hfus	38.05	kJ/mol	Joback Method
hvap	83.42	kJ/mol	Joback Method
log10ws	-5.15		Crippen Method
logp	4.121		Crippen Method
mcvol	264.880	ml/mol	McGowan Method
pc	1752.14	kPa	Joback Method
rinpola	2436.00		NIST Webbook
rinpola	2436.00		NIST Webbook
tb	865.28	K	Joback Method
tc	1081.52	K	Joback Method
tf	529.16	K	Joback Method
vc	0.996	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	771.27	J/molxK	865.28	Joback Method
cpg	784.71	J/molxK	901.32	Joback Method
cpg	796.97	J/molxK	937.36	Joback Method
cpg	808.04	J/molxK	973.40	Joback Method
cpg	817.93	J/molxK	1009.44	Joback Method
cpg	826.64	J/molxK	1045.48	Joback Method
cpg	834.18	J/molxK	1081.52	Joback Method
dvisc	0.0004004	Paxs	529.16	Joback Method

dvisc	0.0002301	Paxs	585.18	Joback Method
dvisc	0.0001457	Paxs	641.20	Joback Method
dvisc	0.0000993	Paxs	697.22	Joback Method
dvisc	0.0000716	Paxs	753.24	Joback Method
dvisc	0.0000541	Paxs	809.26	Joback Method
dvisc	0.0000423	Paxs	865.28	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U393703&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U393703&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>

## Legend

<b>cp<sub>g</sub>:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log<sub>10</sub>ws:</b>	Log <sub>10</sub> of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m<sub>cvol</sub>:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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