

# Benzene, 1-chloro-4-ethynyl-

<b>Other names:</b>	(p-Chlorophenyl)acetylene (4-Chlorophenyl)acetylene p-Ethynylchlorobenzene 1-Chloro-4-ethynylbenzene
<b>Inchi:</b>	InChI=1S/C8H5Cl/c1-2-7-3-5-8(9)6-4-7/h1,3-6H
<b>InchiKey:</b>	LFZJRTMTKGYJRS-UHFFFAOYSA-N
<b>Formula:</b>	C8H5Cl
<b>SMILES:</b>	C#Cc1ccc(Cl)cc1
<b>Mol. weight [g/mol]:</b>	136.58
<b>CAS:</b>	873-73-4

## Physical Properties

Property code	Value	Unit	Source
affp	832.40	kJ/mol	NIST Webbook
basg	801.70	kJ/mol	NIST Webbook
gf	330.40	kJ/mol	Joback Method
hf	292.77	kJ/mol	Joback Method
hfus	17.30	kJ/mol	Joback Method
hvap	40.58	kJ/mol	Joback Method
ie	8.75	eV	NIST Webbook
ie	8.60	eV	NIST Webbook
log10ws	-2.86		Crippen Method
logp	2.321		Crippen Method
mcvol	103.460	ml/mol	McGowan Method
pc	4114.41	kPa	Joback Method
tb	441.65	K	Joback Method
tc	677.32	K	Joback Method
tf	295.75	K	Joback Method
vc	0.387	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	173.63	J/molxK	441.65	Joback Method

cpg	183.38	J/mol×K	480.93	Joback Method
cpg	192.42	J/mol×K	520.21	Joback Method
cpg	200.79	J/mol×K	559.49	Joback Method
cpg	208.54	J/mol×K	598.77	Joback Method
cpg	215.69	J/mol×K	638.04	Joback Method
cpg	222.30	J/mol×K	677.32	Joback Method

## Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C873734&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C873734&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.cheméo.com/doc/models/crippen_log10ws">https://www.cheméo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

## Legend

<b>affp:</b>	Proton affinity
<b>basg:</b>	Gas basicity
<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>ie:</b>	Ionization energy
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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