

# (CH<sub>3</sub>)<sub>2</sub>N-CH=N-(CH<sub>2</sub>)<sub>3</sub>N(CH<sub>3</sub>)<sub>2</sub>

**Inchi:** InChI=1S/C8H19N3/c1-10(2)7-5-6-9-8-11(3)4/h8H,5-7H2,1-4H3  
**InchiKey:** HCQHTPHOWFJCLC-UHFFFAOYSA-N  
**Formula:** C<sub>8</sub>H<sub>19</sub>N<sub>3</sub>  
**SMILES:** CN(C)C=NCCCN(C)C  
**Mol. weight [g/mol]:** 157.26  
**CAS:** 139033-04-8

## Physical Properties

Property code	Value	Unit	Source
affp	1057.70	kJ/mol	NIST Webbook
basg	1010.60	kJ/mol	NIST Webbook
hf	8.83	kJ/mol	Joback Method
hvap	40.80	kJ/mol	Joback Method
log10ws	0.03		Crippen Method
logp	0.528		Crippen Method
mcvol	149.220	ml/mol	McGowan Method
pc	2278.41	kPa	Joback Method
tb	484.00	K	Joback Method
tc	664.40	K	Joback Method

## Sources

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)  
**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>  
**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C139033048&Units=SI>  
**Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci990307I>  
**Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

## Legend

**affp:** Proton affinity

<b>basg:</b>	Gas basicity
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hvac:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

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