

# Chloroacetic acid allyl ester

<b>Other names:</b>	Acetic acid, chloro-, 2-propenyl ester Acetic acid, chloro-, allyl ester Allyl chloroacetate
<b>Inchi:</b>	InChI=1S/C5H7ClO2/c1-2-3-8-5(7)4-6/h2H,1,3-4H2
<b>InchiKey:</b>	VMBJJCDVORDOCF-UHFFFAOYSA-N
<b>Formula:</b>	C5H7ClO2
<b>SMILES:</b>	C=CCOC(=O)CCl
<b>Mol. weight [g/mol]:</b>	134.56
<b>CAS:</b>	2916-14-5

## Physical Properties

Property code	Value	Unit	Source
chl	-2603.00	kJ/mol	NIST Webbook
chl	-2606.00 ± 3.00	kJ/mol	NIST Webbook
gf	-166.79	kJ/mol	Joback Method
hf	-281.64	kJ/mol	Joback Method
hfus	14.41	kJ/mol	Joback Method
hvap	39.59	kJ/mol	Joback Method
log10ws	-0.78		Crippen Method
logp	0.954		Crippen Method
mcvol	96.690	ml/mol	McGowan Method
pc	3686.49	kPa	Joback Method
rinpol	870.00		NIST Webbook
ripol	1429.00		NIST Webbook
tb	418.20	K	NIST Webbook
tc	614.21	K	Joback Method
tf	246.43	K	Joback Method
vc	0.369	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	171.64	J/mol×K	424.20	Joback Method
cpg	206.45	J/mol×K	582.54	Joback Method

cpg	200.06	J/molxK	550.88	Joback Method
cpg	193.40	J/molxK	519.21	Joback Method
cpg	186.44	J/molxK	487.54	Joback Method
cpg	179.18	J/molxK	455.87	Joback Method
cpg	212.54	J/molxK	614.21	Joback Method
dvisc	0.0003131	Paxs	424.20	Joback Method
dvisc	0.0003911	Paxs	394.57	Joback Method
dvisc	0.0005064	Paxs	364.94	Joback Method
dvisc	0.0006863	Paxs	335.32	Joback Method
dvisc	0.0009866	Paxs	305.69	Joback Method
dvisc	0.0015332	Paxs	276.06	Joback Method
dvisc	0.0026492	Paxs	246.43	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C2916145&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C2916145&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chl:</b>	Standard liquid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>ripol:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature

**tf:** Normal melting (fusion) point

**vc:** Critical Volume

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