

# Propyl methylaminodithiocarbamate

<b>Inchi:</b>	InChI=1S/C5H11NS2/c1-3-4-8-5(7)6-2/h3-4H2,1-2H3,(H,6,7)
<b>InchiKey:</b>	KUDJTSQPXXVPAQ-UHFFFAOYSA-N
<b>Formula:</b>	C5H11NS2
<b>SMILES:</b>	CCCSC(=S)NC
<b>Mol. weight [g/mol]:</b>	149.28

## Physical Properties

Property code	Value	Unit	Source
gf	230.79	kJ/mol	Joback Method
hf	95.31	kJ/mol	Joback Method
hfus	22.54	kJ/mol	Joback Method
hvap	46.71	kJ/mol	Joback Method
log10ws	-2.34		Crippen Method
logp	1.634		Crippen Method
mcvol	119.690	ml/mol	McGowan Method
pc	4026.13	kPa	Joback Method
rinpol	1507.00		NIST Webbook
rinpol	1516.00		NIST Webbook
rinpol	1521.00		NIST Webbook
rinpol	1518.00		NIST Webbook
rinpol	1507.00		NIST Webbook
rinpol	1509.00		NIST Webbook
rinpol	1512.00		NIST Webbook
rinpol	1521.00		NIST Webbook
tb	502.79	K	Joback Method
tc	725.26	K	Joback Method
tf	267.44	K	Joback Method
vc	0.441	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	238.58	J/molxK	502.79	Joback Method
cpg	249.00	J/molxK	539.87	Joback Method

cpg	258.76	J/mol×K	576.95	Joback Method
cpg	267.88	J/mol×K	614.03	Joback Method
cpg	276.41	J/mol×K	651.10	Joback Method
cpg	284.39	J/mol×K	688.18	Joback Method
cpg	291.87	J/mol×K	725.26	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=R11228&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=R11228&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m cvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r inpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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