

# Dichlorophen, O,O'-di(cyclopropanecarbonyl)-

<b>Inchi:</b>	InChI=1S/C21H18Cl2O4/c22-16-5-7-18(26-20(24)12-1-2-12)14(10-16)9-15-11-17(23)6-8
<b>InchiKey:</b>	XYIMISYZCZHNOC-UHFFFAOYSA-N
<b>Formula:</b>	C21H18Cl2O4
<b>SMILES:</b>	O=C(Oc1ccc(Cl)cc1Cc1cc(Cl)ccc1OC(=O)C1CC1)C1CC1
<b>Mol. weight [g/mol]:</b>	405.27

## Physical Properties

Property code	Value	Unit	Source
gf	-57.96	kJ/mol	Joback Method
hf	-425.07	kJ/mol	Joback Method
hfus	46.91	kJ/mol	Joback Method
hvap	96.45	kJ/mol	Joback Method
log10ws	-6.55		Crippen Method
logp	5.215		Crippen Method
mvol	276.870	ml/mol	McGowan Method
pc	1804.63	kPa	Joback Method
rinpol	2920.00		NIST Webbook
rinpol	2920.00		NIST Webbook
tb	994.08	K	Joback Method
tc	1244.22	K	Joback Method
tf	669.39	K	Joback Method
vc	1.056	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	842.77	J/molxK	994.08	Joback Method
cpg	855.08	J/molxK	1035.77	Joback Method
cpg	866.46	J/molxK	1077.46	Joback Method
cpg	877.02	J/molxK	1119.15	Joback Method
cpg	886.90	J/molxK	1160.84	Joback Method
cpg	896.22	J/molxK	1202.53	Joback Method
cpg	905.09	J/molxK	1244.22	Joback Method
dvisc	0.0013016	Paxs	669.39	Joback Method

dvisc	0.0010183	Paxs	723.50	Joback Method
dvisc	0.0008243	Paxs	777.62	Joback Method
dvisc	0.0006859	Paxs	831.73	Joback Method
dvisc	0.0005837	Paxs	885.85	Joback Method
dvisc	0.0005060	Paxs	939.96	Joback Method
dvisc	0.0004455	Paxs	994.08	Joback Method

## Sources

<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=U354668&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=U354668&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpolar:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/115-869-0/Dichlorophen-O-O-di-cyclopropanecarbonyl.pdf>

Generated by Cheméo on 2024-04-28 21:39:28.712609842 +0000 UTC m=+16629617.633187158.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.