

Succinic acid, 3-chlorophenyl 2,2,3,3,3-pentafluoropropyl ester

Inchi: InChI=1S/C13H10ClF5O4/c14-8-2-1-3-9(6-8)23-11(21)5-4-10(20)22-7-12(15,16)13(17,18)
InchiKey: IFVNNPJCJUPTFP-UHFFFAOYSA-N
Formula: C13H10ClF5O4
SMILES: O=C(CCC(=O)Oc1cccc(Cl)c1)OCC(F)(F)C(F)(F)F
Mol. weight [g/mol]: 360.66

Physical Properties

Property code	Value	Unit	Source
gf	-1286.78	kJ/mol	Joback Method
hf	-1589.98	kJ/mol	Joback Method
hfus	33.42	kJ/mol	Joback Method
hvap	63.49	kJ/mol	Joback Method
log10ws	-4.40		Crippen Method
logp	3.766		Crippen Method
mvol	206.240	ml/mol	McGowan Method
pc	1920.30	kPa	Joback Method
rinpol	1678.00		NIST Webbook
rinpol	1678.00		NIST Webbook
tb	708.40	K	Joback Method
tc	901.23	K	Joback Method
tf	457.24	K	Joback Method
vc	0.821	m ³ /kmol	Joback Method

Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	558.24	J/mol×K	708.40	Joback Method
cpg	569.09	J/mol×K	740.54	Joback Method
cpg	579.11	J/mol×K	772.68	Joback Method
cpg	588.35	J/mol×K	804.81	Joback Method
cpg	596.86	J/mol×K	836.95	Joback Method
cpg	604.66	J/mol×K	869.09	Joback Method
cpg	611.81	J/mol×K	901.23	Joback Method

Sources

Crippen Method:	http://pubs.acs.org/doi/abs/10.1021/ci9903071
Crippen Method:	https://www.cheméo.com/doc/models/crippen_log10ws
Joback Method:	https://en.wikipedia.org/wiki/Joback_method
McGowan Method:	http://link.springer.com/article/10.1007/BF02311772
NIST Webbook:	http://webbook.nist.gov/cgi/cbook.cgi?ID=U390871&Units=SI

Legend

cpg:	Ideal gas heat capacity
gf:	Standard Gibbs free energy of formation
hf:	Enthalpy of formation at standard conditions
hfus:	Enthalpy of fusion at standard conditions
h vap:	Enthalpy of vaporization at standard conditions
log10ws:	Log10 of Water solubility in mol/l
logp:	Octanol/Water partition coefficient
mcvol:	McGowan's characteristic volume
pc:	Critical Pressure
r in pol:	Non-polar retention indices
tb:	Normal Boiling Point Temperature
tc:	Critical Temperature
tf:	Normal melting (fusion) point
vc:	Critical Volume

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