

# Carbaril

## Other names:

.alpha.-naphthalenyl methylcarbamate  
.alpha.-naphthyl methylcarbamate  
1-Naphthalenol, 1-(N-methylcarbamate)  
1-Naphthalenol, methylcarbamate  
1-Naphthol N-methylcarbamate  
1-Naphthyl N-methylcarbamate  
1-Naphthyl N-methylcarbamateacid O,O-diethyl ester  
1-Naphthyl methylcarbamate  
1-naphthalenyl methylcarbamate  
Adios  
Arilat  
Arilate  
Arylam  
Atoxan  
Bercema NMC50  
Caprolin  
Carbamic acid, N-methyl,1-naphthyl ester  
Carbamic acid, N-methyl-1-naphthyl-  
Carbamic acid, methyl-, 1-naphthyl ester  
Carbamine  
Carbarilo  
Carbarilum  
Carbaryl  
Carbatox  
Carbatox 75  
Carbatox-60  
Carbavur  
Carbomate  
Carpolin  
Carylderm  
Clinicide  
Compound 7744  
Crag sevin  
Denapon  
Derbac  
Dicarbam  
Dicarbament 23,969  
Dyna-carbyl  
ENT-23969  
Experimental Insecticide 7744

Gamonil  
Germain's  
Hexavin  
Karbaryl  
Karbapray  
Karbatox  
Karbatox 75  
Karbosep  
Laivin  
Menaphtam  
Methylcarbamate, 1-naphthalenol  
Methylcarbamic acid, 1-naphthyl ester  
Monsur  
Mugan  
Murvin  
N-Methyl-1-naftyl-carbamaat  
N-Methyl-1-naphthyl carbamate  
N-Methyl-1-naphthyl-carbamat  
N-Methyl-«alpha»-Naphthylcarbamate  
N-Methyl-«alpha»-naphthylurethan  
N-Methyl-«alpha»-Naphthylcarbamate  
N-Methyl-«alpha»-naphthylurethan  
N-Methylcarbamate de 1-naphtyle  
N-Metil-1-naftil-carbammato  
NAC  
NMC 50  
OMS-29  
Oltitox  
Panam  
Pomex  
Prosevor 85  
Ravyon  
SOK  
Seffein  
Septene  
Sevimol  
Sevin  
Sevin 4  
Sevin SL  
Sewin  
Suleo  
Thinsec  
Tornado

Tricarnam  
 UC 7744  
 Union Carbide 7744  
 Vetox  
 Vetox 85  
 Vioxan  
 naphthalen-1-yl N-methylcarbamate  
 «alpha»-Naftyl-N-methylkarbamat  
 «alpha»-Naphthalenyl methylcarbamate  
 «alpha»-Naphthyl N-methylcarbamate  
 «alpha»-Naphthyl methylcarbamate  
 Â«alphaÂ»-Naftyl-N-methylkarbamat  
 Â«alphaÂ»-Naphthalenyl methylcarbamate  
 Â«alphaÂ»-Naphthyl N-methylcarbamate  
 Â«alphaÂ»-Naphthyl methylcarbamate  
**Inchi:** InChI=1S/C12H11NO2/c1-13-12(14)15-11-8-4-6-9-5-2-3-7-10(9)11/h2-8H,1H3,(H,13,14)  
**InchiKey:** CVXBEEKQHEXEN-UHFFFAOYSA-N  
**Formula:** C12H11NO2  
**SMILES:** CNC(=O)Oc1cccc2ccccc12  
**Mol. weight [g/mol]:** 201.22  
**CAS:** 63-25-2

## Physical Properties

Property code	Value	Unit	Source
gf	115.06	kJ/mol	Joback Method
hf	-66.21	kJ/mol	Joback Method
hfus	25.39	kJ/mol	Joback Method
hvap	62.48	kJ/mol	Joback Method
log10ws	-2.94		Aqueous Solubility Prediction Method
log10ws	-3.22		Estimated Solubility Method
logp	2.558		Crippen Method
mcvol	154.140	ml/mol	McGowan Method
pc	3265.31	kPa	Joback Method
rinpol	1915.00		NIST Webbook
rinpol	1903.00		NIST Webbook
rinpol	1903.00		NIST Webbook
rinpol	1912.00		NIST Webbook
rinpol	1865.00		NIST Webbook

rinpol	1871.00		NIST Webbook
rinpol	1910.00		NIST Webbook
ripol	2826.00		NIST Webbook
tb	651.06	K	Joback Method
tc	884.28	K	Joback Method
tf	415.50 ± 0.20	K	NIST Webbook
tf	415.90 ± 0.20	K	NIST Webbook
tf	416.89 ± 0.20	K	NIST Webbook
tf	415.90	K	Aqueous Solubility Prediction Method
vc	0.581	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	439.72	J/mol×K	845.41	Joback Method
cpg	384.13	J/mol×K	651.06	Joback Method
cpg	397.06	J/mol×K	689.93	Joback Method
cpg	409.02	J/mol×K	728.80	Joback Method
cpg	420.08	J/mol×K	767.67	Joback Method
cpg	430.30	J/mol×K	806.54	Joback Method
cpg	448.41	J/mol×K	884.28	Joback Method
hfust	24.51	kJ/mol	416.30	NIST Webbook

## Sources

**NIST Webbook:**

<http://webbook.nist.gov/cgi/cbook.cgi?ID=C63252&Units=SI>

**Crippen Method:**

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

**Thermodynamics of dissolution for 1-naphtol N-methylcarbamate in pure, ionized and Mediterranean sea water by dynamic saturation method;**

<https://www.doi.org/10.1016/j.fluid.2007.11.012>

[https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)

**Aqueous Solubility Prediction Method:**

<http://onschallenge.wikispaces.com/file/view/AqueousDataset002.xlsx/351826032/AqueousDataset002.xlsx>

**Estimated Solubility Method:**

[http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl\\_file/ci034243xsi20040112\\_053635.txt](http://pubs.acs.org/doi/suppl/10.1021/ci034243x/suppl_file/ci034243xsi20040112_053635.txt)

**McGowan Method:**

<http://link.springer.com/article/10.1007/BF02311772>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpola:</b>	Non-polar retention indices
<b>ripola:</b>	Polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.cheméo.com/cid/11-703-8/Carbaril.pdf>

Generated by Cheméo on 2024-04-18 15:07:23.348991654 +0000 UTC m=+15742092.269568977.

Cheméo (<https://www.cheméo.com>) is the biggest free database of chemical and physical data for the process industry.