

# Butanedioic acid, methyl-

<b>Other names:</b>	Succinic acid, methyl- Methylsuccinic acid Pyrotartaric acid 1,2-Propanedicarboxylic acid Methylbutanedioic acid 2-Methylsuccinic acid 2-Methylbutanedioic acid
<b>Inchi:</b>	InChI=1S/C5H8O4/c1-3(5(8)9)2-4(6)7/h3H,2H2,1H3,(H,6,7)(H,8,9)
<b>InchiKey:</b>	WXUAQHNMJWJLTG-UHFFFAOYSA-N
<b>Formula:</b>	C5H8O4
<b>SMILES:</b>	CC(CC(=O)O)C(=O)O
<b>Mol. weight [g/mol]:</b>	132.11
<b>CAS:</b>	498-21-5

## Physical Properties

Property code	Value	Unit	Source
chs	-2152.70 ± 1.00	kJ/mol	NIST Webbook
gf	-542.70	kJ/mol	Joback Method
hf	-681.43	kJ/mol	Joback Method
hfus	16.56	kJ/mol	Joback Method
hvap	73.19	kJ/mol	Joback Method
log10ws	0.13		Crippen Method
logp	0.182		Crippen Method
mcvol	96.190	ml/mol	McGowan Method
pc	5251.00	kPa	Joback Method
tb	380.65 ± 2.00	K	NIST Webbook
tc	783.09	K	Joback Method
tf	352.61	K	Joback Method
vc	0.359	m3/kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	232.92	J/mol×K	605.46	Joback Method

cpg	239.21	J/mol×K	635.07	Joback Method
cpg	245.20	J/mol×K	664.67	Joback Method
cpg	250.87	J/mol×K	694.28	Joback Method
cpg	256.24	J/mol×K	723.88	Joback Method
cpg	261.32	J/mol×K	753.49	Joback Method
cpg	266.11	J/mol×K	783.09	Joback Method
cps	199.60	J/mol×K	323.00	NIST Webbook
dvisc	0.0123933	Paxs	352.61	Joback Method
dvisc	0.0029414	Paxs	394.75	Joback Method
dvisc	0.0009213	Paxs	436.89	Joback Method
dvisc	0.0003540	Paxs	479.04	Joback Method
dvisc	0.0001588	Paxs	521.18	Joback Method
dvisc	0.0000803	Paxs	563.32	Joback Method
dvisc	0.0000446	Paxs	605.46	Joback Method

## Sources

<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C498215&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C498215&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>

## Legend

<b>chs:</b>	Standard solid enthalpy of combustion
<b>cpg:</b>	Ideal gas heat capacity
<b>cps:</b>	Solid phase heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature

**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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