

# perbenzoic acid

<b>Inchi:</b>	InChI=1S/C7H6O3/c8-7(10-9)6-4-2-1-3-5-6/h1-5,9H
<b>InchiKey:</b>	XCRBXWCUXJNEFX-UHFFFAOYSA-N
<b>Formula:</b>	C7H6O3
<b>SMILES:</b>	O=C(OO)c1ccccc1
<b>Mol. weight [g/mol]:</b>	138.12
<b>CAS:</b>	93-59-4

## Physical Properties

Property code	Value	Unit	Source
gf	-250.27	kJ/mol	Joback Method
hf	-348.31	kJ/mol	Joback Method
hfs	-367.00 ± 13.00	kJ/mol	NIST Webbook
hfus	14.80	kJ/mol	Joback Method
hvap	59.29	kJ/mol	Joback Method
log10ws	-1.61		Crippen Method
logp	1.316		Crippen Method
mcvol	99.040	ml/mol	McGowan Method
pc	4952.36	kPa	Joback Method
tb	554.71	K	Joback Method
tc	762.15	K	Joback Method
tf	328.05	K	Joback Method
vc	0.362	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	214.80	J/mol×K	554.71	Joback Method
cpg	223.27	J/mol×K	589.28	Joback Method
cpg	231.24	J/mol×K	623.86	Joback Method
cpg	238.70	J/mol×K	658.43	Joback Method
cpg	245.69	J/mol×K	693.00	Joback Method
cpg	252.19	J/mol×K	727.58	Joback Method
cpg	258.24	J/mol×K	762.15	Joback Method
dvisc	0.0052845	Paxs	328.05	Joback Method

dvisc	0.0019974	Paxs	365.83	Joback Method
dvisc	0.0009058	Paxs	403.60	Joback Method
dvisc	0.0004703	Paxs	441.38	Joback Method
dvisc	0.0002708	Paxs	479.16	Joback Method
dvisc	0.0001690	Paxs	516.93	Joback Method
dvisc	0.0001125	Paxs	554.71	Joback Method

## Sources

<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C93594&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C93594&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci9903071">http://pubs.acs.org/doi/abs/10.1021/ci9903071</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>
<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfs:</b>	Solid phase enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mcvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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