

# Phenol, 3-pentadecyl-

<b>Other names:</b>	3-n-Pentadecylphenol 3-pentadecylphenol Anacardol, tetrahydro- Cardolite NC-507 Cardolite NC-510 CyclogallipharaoI Hydroginkgol NSC 9781 hydrocardanol m-pentadecylphenol phenol, m-pentadecyl- tetrahydroanacardol
<b>Inchi:</b>	InChI=1S/C21H36O/c1-2-3-4-5-6-7-8-9-10-11-12-13-14-16-20-17-15-18-21(22)19-20/h15
<b>InchiKey:</b>	PTFIPECGHSYQNR-UHFFFAOYSA-N
<b>Formula:</b>	C21H36O
<b>SMILES:</b>	CCCCCCCCCCCCCCCc1cccc(O)c1
<b>Mol. weight [g/mol]:</b>	304.51
<b>CAS:</b>	501-24-6

## Physical Properties

Property code	Value	Unit	Source
gf	83.73	kJ/mol	Joback Method
hf	-417.55	kJ/mol	Joback Method
hfus	38.09	kJ/mol	Solubilities of 3-Pentadecylphenol in Ethanol, 1-Butanol, Toluene, Acetone, Tetrachloromethane, and Ethyl Acetate
hvap	77.63	kJ/mol	Joback Method
log10ws	-7.26		Crippen Method
logp	7.026		Crippen Method
mcvol	288.860	ml/mol	McGowan Method
pc	1318.48	kPa	Joback Method
rinpol	2499.80		NIST Webbook
rinpol	2499.80		NIST Webbook
tb	787.18	K	Joback Method
tc	980.56	K	Joback Method

tf	325.20 ± 1.50	K	NIST Webbook
vc	1.069	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	896.46	J/mol×K	787.18	Joback Method
cpg	915.30	J/mol×K	819.41	Joback Method
cpg	933.25	J/mol×K	851.64	Joback Method
cpg	950.39	J/mol×K	883.87	Joback Method
cpg	966.79	J/mol×K	916.10	Joback Method
cpg	982.54	J/mol×K	948.33	Joback Method
cpg	997.72	J/mol×K	980.56	Joback Method
dvisc	0.0003779	Paxs	464.57	Joback Method
dvisc	0.0001284	Paxs	518.34	Joback Method
dvisc	0.0000534	Paxs	572.11	Joback Method
dvisc	0.0000259	Paxs	625.88	Joback Method
dvisc	0.0000140	Paxs	679.64	Joback Method
dvisc	0.0000083	Paxs	733.41	Joback Method
dvisc	0.0000053	Paxs	787.18	Joback Method
hfust	38.09	kJ/mol	322.40	NIST Webbook

## Pressure Dependent Properties

Property code	Value	Unit	Pressure [kPa]	Source
tbrp	465.70	K	0.10	NIST Webbook
tbrp	470.20	K	0.10	NIST Webbook

## Sources

- McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>
- NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C501246&Units=SI>
- Crippen Method:** <http://pubs.acs.org/doi/abs/10.1021/ci9903071>
- Crippen Method:** [https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)
- Solubilities of 3-Pentadecylphenol in Ethanol, 1-Butanol, Toluene, Acetone, Tetrachloromethane, and Ethyl Acetate:** <https://www.doi.org/10.1021/je900346t>

## Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>dvisc:</b>	Dynamic viscosity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>hfust:</b>	Enthalpy of fusion at a given temperature
<b>hvap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>mccvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>rinpol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tbrp:</b>	Boiling point at reduced pressure
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

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