

# Cyclopropane, 1,1,2,2-tetramethyl-

<b>Other names:</b>	1,1,2,2-Tetramethylcyclopropane
<b>Inchi:</b>	InChI=1S/C7H14/c1-6(2)5-7(6,3)4/h5H2,1-4H3
<b>InchiKey:</b>	JCHUCGKEGUAHEH-UHFFFAOYSA-N
<b>Formula:</b>	C7H14
<b>SMILES:</b>	CC1(C)CC1(C)C
<b>Mol. weight [g/mol]:</b>	98.19
<b>CAS:</b>	4127-47-3

## Physical Properties

Property code	Value	Unit	Source
chl	-4635.62 ± 0.84	kJ/mol	NIST Webbook
gf	50.12	kJ/mol	Joback Method
hf	-104.87	kJ/mol	Joback Method
hfl	-119.80 ± 0.92	kJ/mol	NIST Webbook
hfus	0.50	kJ/mol	Joback Method
hvap	28.48	kJ/mol	Joback Method
log10ws	-2.16		Crippen Method
logp	2.442		Crippen Method
mcvol	98.630	ml/mol	McGowan Method
pc	3448.03	kPa	Joback Method
rinpol	621.20		NIST Webbook
rinpol	621.00		NIST Webbook
rinpol	621.00		NIST Webbook
rinpol	622.00		NIST Webbook
rinpol	620.00		NIST Webbook
rinpol	622.00		NIST Webbook
rinpol	621.00		NIST Webbook
rinpol	618.00		NIST Webbook
rinpol	621.00		NIST Webbook
rinpol	622.00		NIST Webbook
rinpol	621.00		NIST Webbook
rinpol	625.00		NIST Webbook
rinpol	618.00		NIST Webbook
rinpol	620.00		NIST Webbook
rinpol	622.00		NIST Webbook
rinpol	618.00		NIST Webbook
tb	349.00 ± 0.20	K	NIST Webbook

tb	349.00 ± 0.30	K	NIST Webbook
tb	349.00 ± 0.30	K	NIST Webbook
tb	349.04 ± 0.30	K	NIST Webbook
tb	349.00 ± 0.20	K	NIST Webbook
tc	556.59	K	Joback Method
tf	230.15	K	Joback Method
vc	0.380	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	177.90	J/mol×K	362.11	Joback Method
cpg	193.91	J/mol×K	394.52	Joback Method
cpg	208.42	J/mol×K	426.94	Joback Method
cpg	221.59	J/mol×K	459.35	Joback Method
cpg	233.56	J/mol×K	491.76	Joback Method
cpg	244.49	J/mol×K	524.18	Joback Method
cpg	254.53	J/mol×K	556.59	Joback Method

## Correlations

Information	Value
Property code	pvap
Equation	$\ln(P_{vp}) = A + B/(T + C)$
Coeff. A	1.73089e+01
Coeff. B	-3.89273e+03
Coeff. C	-4.22570e+01
Temperature range (K), min.	270.96
Temperature range (K), max.	366.72

## Sources

The Yaws Handbook of Vapor  
Pressure:  
Crippen Method:  
Crippen Method:

<https://www.sciencedirect.com/book/9780128029992/the-yaws-handbook-of-vapor-pressure>

<http://pubs.acs.org/doi/abs/10.1021/ci9903071>

[https://www.chemeo.com/doc/models/crippen\\_log10ws](https://www.chemeo.com/doc/models/crippen_log10ws)

**Joback Method:** [https://en.wikipedia.org/wiki/Joback\\_method](https://en.wikipedia.org/wiki/Joback_method)  
**McGowan Method:** <http://link.springer.com/article/10.1007/BF02311772>  
**NIST Webbook:** <http://webbook.nist.gov/cgi/cbook.cgi?ID=C4127473&Units=SI>

## Legend

**chl:** Standard liquid enthalpy of combustion  
**cpg:** Ideal gas heat capacity  
**gf:** Standard Gibbs free energy of formation  
**hf:** Enthalpy of formation at standard conditions  
**hfl:** Liquid phase enthalpy of formation at standard conditions  
**hfus:** Enthalpy of fusion at standard conditions  
**hvap:** Enthalpy of vaporization at standard conditions  
**log10ws:** Log10 of Water solubility in mol/l  
**logp:** Octanol/Water partition coefficient  
**mcvol:** McGowan's characteristic volume  
**pc:** Critical Pressure  
**pvap:** Vapor pressure  
**rinpol:** Non-polar retention indices  
**tb:** Normal Boiling Point Temperature  
**tc:** Critical Temperature  
**tf:** Normal melting (fusion) point  
**vc:** Critical Volume

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