

# 2,3,4,5,6-Pentafluoro-N-(4-fluorophenyl)benzamid

<b>Other names:</b>	Benzamide, N-(4-fluorophenyl)-2,3,4,5,6-pentafluoro-
<b>Inchi:</b>	InChI=1S/C13H5F6NO/c14-5-1-3-6(4-2-5)20-13(21)7-8(15)10(17)12(19)11(18)9(7)16/h1
<b>InchiKey:</b>	NXSQZVGPXRXGT-UHFFFAOYSA-N
<b>Formula:</b>	C13H5F6NO
<b>SMILES:</b>	O=C(Nc1ccc(F)cc1)c1c(F)c(F)c(F)c(F)c1F
<b>Mol. weight [g/mol]:</b>	305.18
<b>CAS:</b>	209109-65-9

## Physical Properties

Property code	Value	Unit	Source
gf	-982.77	kJ/mol	Joback Method
hf	-1143.18	kJ/mol	Joback Method
hfus	40.35	kJ/mol	Joback Method
hvap	61.34	kJ/mol	Joback Method
log10ws	-5.44		Crippen Method
logp	3.773		Crippen Method
mcvol	168.680	ml/mol	McGowan Method
pc	2293.71	kPa	Joback Method
rinpole	1716.00		NIST Webbook
tb	679.74	K	Joback Method
tc	875.59	K	Joback Method
tf	470.36	K	Joback Method
vc	0.697	m <sup>3</sup> /kmol	Joback Method

## Temperature Dependent Properties

Property code	Value	Unit	Temperature [K]	Source
cpg	431.93	J/molxK	679.74	Joback Method
cpg	441.80	J/molxK	712.38	Joback Method
cpg	451.04	J/molxK	745.02	Joback Method
cpg	459.66	J/molxK	777.66	Joback Method
cpg	467.66	J/molxK	810.30	Joback Method
cpg	475.08	J/molxK	842.95	Joback Method
cpg	481.93	J/molxK	875.59	Joback Method

# Sources

<b>McGowan Method:</b>	<a href="http://link.springer.com/article/10.1007/BF02311772">http://link.springer.com/article/10.1007/BF02311772</a>
<b>NIST Webbook:</b>	<a href="http://webbook.nist.gov/cgi/cbook.cgi?ID=C209109659&amp;Units=SI">http://webbook.nist.gov/cgi/cbook.cgi?ID=C209109659&amp;Units=SI</a>
<b>Crippen Method:</b>	<a href="http://pubs.acs.org/doi/abs/10.1021/ci990307l">http://pubs.acs.org/doi/abs/10.1021/ci990307l</a>
<b>Crippen Method:</b>	<a href="https://www.chemeo.com/doc/models/crippen_log10ws">https://www.chemeo.com/doc/models/crippen_log10ws</a>
<b>Joback Method:</b>	<a href="https://en.wikipedia.org/wiki/Joback_method">https://en.wikipedia.org/wiki/Joback_method</a>

# Legend

<b>cpg:</b>	Ideal gas heat capacity
<b>gf:</b>	Standard Gibbs free energy of formation
<b>hf:</b>	Enthalpy of formation at standard conditions
<b>hfus:</b>	Enthalpy of fusion at standard conditions
<b>h vap:</b>	Enthalpy of vaporization at standard conditions
<b>log10ws:</b>	Log10 of Water solubility in mol/l
<b>logp:</b>	Octanol/Water partition coefficient
<b>m cvol:</b>	McGowan's characteristic volume
<b>pc:</b>	Critical Pressure
<b>r in pol:</b>	Non-polar retention indices
<b>tb:</b>	Normal Boiling Point Temperature
<b>tc:</b>	Critical Temperature
<b>tf:</b>	Normal melting (fusion) point
<b>vc:</b>	Critical Volume

Latest version available from:

<https://www.chemeo.com/cid/10-170-1/2-3-4-5-6-Pentafluoro-N-4-fluorophenyl-benzamide.pdf>

Generated by Cheméo on 2024-04-20 06:59:36.389563742 +0000 UTC m=+15885625.310141053.

Cheméo (<https://www.chemeo.com>) is the biggest free database of chemical and physical data for the process industry.